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Earth, Planets and Space

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Water distribution in quartz schists of the Sanbagawa Metamorphic Belt, Japan: infrared spectroscopic mapping and comparison of the calibrations proposed for determining water contents

Jun-ichi Fukuda[1,2,3*] and Ichiko Shimizu[1,4]

Abstract

We evaluated water distributions in deformed quartz in schists along the Asemi River, Central Shikoku, in the Sanbagawa Metamorphic Belt, Japan, using infrared spectroscopic (IR) mapping. The water trapped in quartz as molecular H_2O showed a broad IR absorption band at 2800–3750 cm^{-1}. A necessary step before assessing the quartz water content was to evaluate and compare six previously proposed IR calibrations in terms of the molar absorption coefficients of H_2O (L/mol H_2O cm^2). The coefficients vary from 24,100 to 89,000 L/mol H_2O cm^2, and the values of the coefficients show a rough increase with increasing component of structural –OH in the IR spectra. We used Paterson's calibration, which does not require input regarding the mineral species, but which was modified in his paper for measurements of molecular H_2O in quartz. The absorption coefficient is 38,000 L/mol H_2O cm^2. IR mapping was performed on Sanbagawa metamorphic rocks with increasing grades of metamorphism, where the mean grain size of quartz increases from ~40 to ~120 μm. The absorption bands that are only from the quartz can be distinguished on the basis of microstructural observations and the corresponding mapping results. The IR spectra of quartz commonly show dominant molecular H_2O bands at 2800–3750 cm^{-1} with no additional bands associated with crystalline –OH when only quartz is measured. The water contents of quartz in all our samples were 40–310 ppm, and these values are about one-third of previously reported values measured using point analyses with the unified Paterson's calibration. This difference seems to reflect the incorporation of phyllosilicates in previous measurements that showed a broad band around 3600 cm^{-1}. The lowest and highest water contents in our quartz samples are associated with intragranular water and grain boundary water, respectively. We estimated the grain boundary widths to be at most ~10 nm on the basis of the water contents at grain boundaries.

Keywords: Quartz, Water content, Infrared spectroscopy, Calibration, Sanbagawa Metamorphic Belt

Introduction

Quartz is widespread in the Earth's crust and is a control on rheology. It is well known that water affects the plastic deformation of quartz, as confirmed by many experimental works (e.g., Griggs and Blacic 1965; Griggs 1967; Parrish et al. 1976; Jaoul et al. 1984; Kronenberg and Tullis 1984; Koch et al. 1989; Post et al. 1996; Chernak et al. 2009; Holyoke and Kronenberg 2013; Fukuda et al. 2018) and theoretical models (Fukuda and Shimizu 2017). However, the quantitative relationship between water content and the plastic deformation of quartz has not been established, even though a rough link between the plastic weakness of quartz and water content is well known. In addition to studies of quartz in deformation

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experiments, the water contents of naturally deformed quartz have also been measured (Kronenberg and Wolf 1990; Kronenberg et al. 1990; Kronenberg 1994; Nakashima et al. 1995; Niimi et al. 1999; Muto et al. 2004, 2005; Gleason and DeSisto 2008; Menegon et al. 2011; Fukuda 2012; Finch et al. 2016; Kilian et al. 2016; Kronenberg et al. 2017), although it is still difficult to relate the measured water content to the development of plastic deformation in nature. To measure the water contents in quartz and to assess the nature of the water species (i.e., as molecular H_2O or –OH coupled with cations), infrared (IR) spectroscopy is used. IR mapping measurements can be correlated with the textures of the sample, and they are useful for evaluating the distribution of water in the area of analysis (e.g., Ito and Nakashima 2002; Fukuda 2012; Fukuda et al. 2012; Kronenberg et al. 2017).

In this paper we present the results of IR mapping measurements to show how water is distributed in and around deformed quartz in the Sanbagawa (or Sambagawa) Metamorphic Belt along the Asemi River in Shikoku, Japan, where the deformed quartz represents a variety of metamorphic conditions. The water contents in quartz aggregates in this area were also measured by Nakashima et al. (1995), who developed an IR calibration to determine the water contents in the quartz, and here we will compare their scheme with other calibrations. Nakashima et al. (1995) also performed point IR analyses, but information on the textures of the samples was not linked to the IR results. In this paper we discuss sample textures and water contents measured by IR mapping. We summarize various IR calibrations that have been proposed, and we compare them in terms of the absorption coefficients using the same unit (L/mol H_2O cm^2).

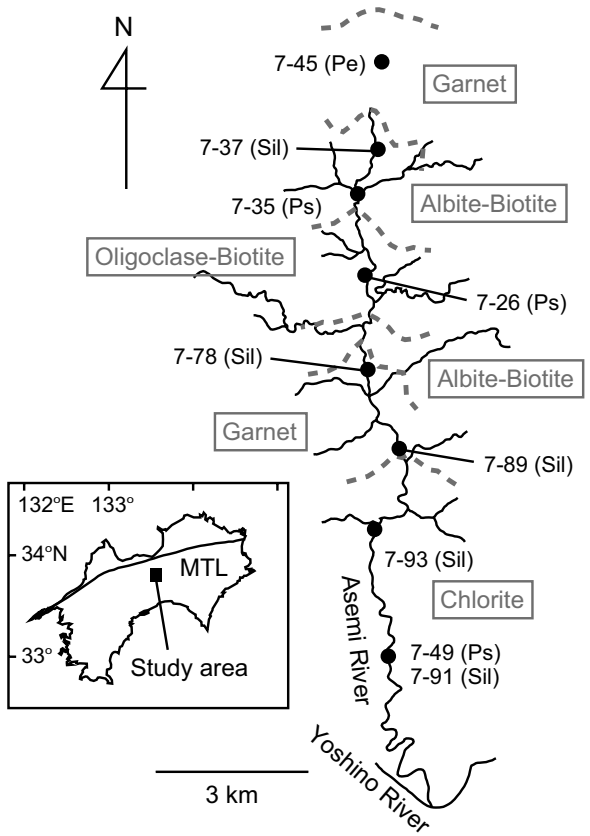


Fig. 1 Sample locations along the Asemi River, also showing metamorphic zones. The inset shows the location of the study area in Shikoku, SW Japan. MTL = Median Tectonic Line. The letters in parentheses after the sample numbers indicate rock types: *Pe* pelitic, *Ps* psammitic, *Sil* siliceous. The metamorphic zones are based on Higashino (1990)

Samples

Deformed metasedimentary rocks from the Sanbagawa Metamorphic Belt were collected along the Asemi River in Shikoku, Japan. The samples increase in metamorphic grade from the chlorite to the biotite zone along the river from south to north (e.g., Banno et al. 1978; Itaya 1981; Higashino 1990). Enami et al. (1994) estimated the pressures and temperatures of the chlorite, garnet, and biotite zones in this region to have been 5.5–6.5 kbar and < 360 °C, 7.0–8.5 kbar and 440 ± 15 °C, and 8.0–9.5 kbar and 520 ± 25 °C, respectively. Higashino (1990) further divided the biotite zone into albite–biotite and oligoclase–biotite zones (Fig. 1).

For our study, three rock types (siliceous, pelitic, and psammitic schists) were used for the IR analyses. Figure 2 shows examples of their textures in samples from the three metamorphic zones (No. 7-93, chlorite zone; No. 7-45, garnet zone; and No. 7-26, oligoclase–biotite zone). Each sample was cut parallel to the lineation and perpendicular to the foliation (i.e., XZ sections). The samples consist mainly of layers of quartz and albite (or oligoclase) along with patches of phyllosilicate minerals such as muscovite and chlorite, with biotite also present in the highest grade zone. The phyllosilicates are sometimes concentrated in layers, and porphyroblasts of albite are locally present. The accessory minerals include hematite, garnet, piemontite, calcite, and apatite. Relic detrital grains of feldspar and quartz form porphyroclasts in the psammitic schist samples of the chlorite zone.

Tagami and Takeshita (1998) and Yagi and Takeshita (2002) reported type-I and type-II cross-girdles (Lister 1977) for quartz *c*-axes fabrics in this area. Yagi and Takeshita (2002) measured the grain sizes of quartz as the diameters of circles equivalent to the areas of the approximately elliptical grains. They reported an increase in the average grain size of the recrystallized quartz from ~ 20 to ~ 180 μm with increasing metamorphic grade.

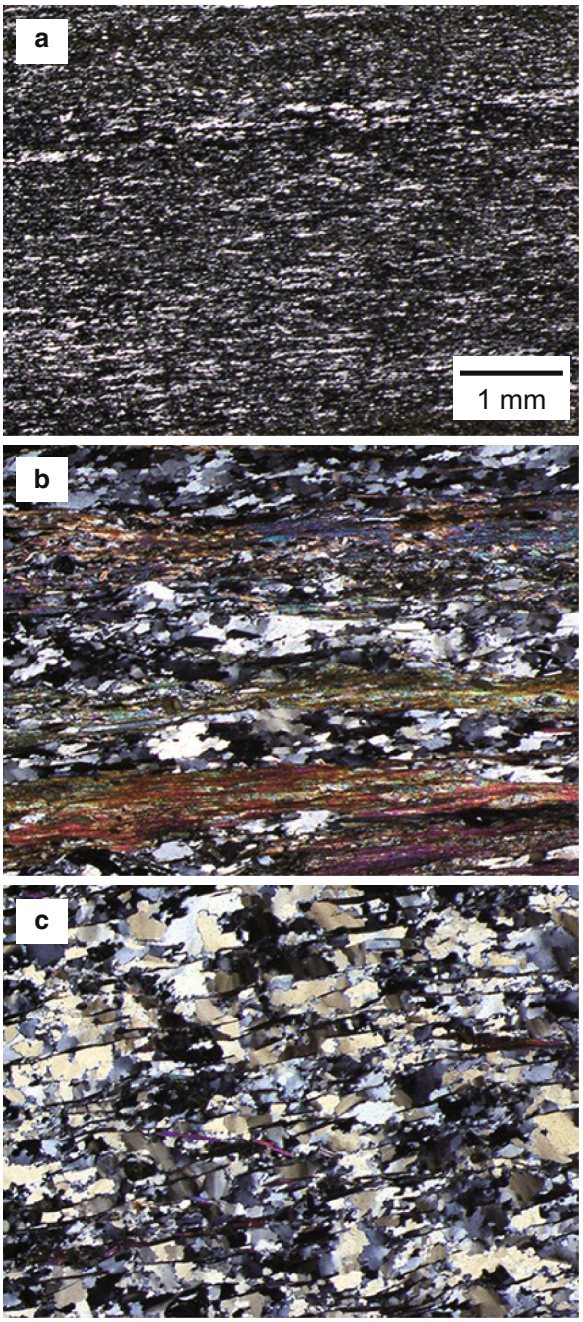


Fig. 2 Optical photomicrographs of the metamorphic rock specimens from different metamorphic zones (cross-polarized light). See Fig. 1 for the locations. **a** Sample No. 7-93 (chlorite zone). **b** Sample No. 7-45 (garnet zone). **c** Sample No. 7-26 (oligoclase–biotite zone). The scale is the same in all images

Masuda (1982) classified grain morphologies of recrystallized quartz along the Asemi River route into P- and S-types, which were originally defined for experimentally deformed and recrystallized samples by Masuda and Fujimura (1981). Yagi and Takeshita (2002) categorized the recrystallized microstructures of quartz as type-I, type-II, or type-III, following the definitions of Hara et al. (1988, 1992). The type-I quartz microstructure is characterized by highly elongate quartz grains with aspect ratios of ≥ 4.0 that display undulatory extinction as well as subgrain formation in and around the host grains, and type-I corresponds to the S-type microstructure of Masuda and Fujimura (1981). The type-II microstructure is characterized by quartz grains that have aspect ratios of ≤ 4.0, grain boundaries that are straight or lobate, and no clear relationships between the host grains and subgrains, and type-II corresponds to the P-type microstructure of Masuda and Fujimura (1981). The type-III microstructure is characterized by quartz grains that have a shape fabric that is oblique to the foliation defined by the alignment of phyllosilicates. Based on these definitions, the quartz microstructures of our samples (Fig. 2) can be classified into type-I or type-II (No. 7-93), type-II or type-III (No. 7-45), and type-III (No. 7-26). The average grain sizes of the quartz, using the method of Yagi and Takeshita (2002), range from ~40 to ~120 μm.

Analytical methods

We measured the IR spectra of the quartz in our schist samples, together with the spectra of other minerals for comparison, and we focused on the distributions of water. We also prepared a specimen of chalcedony from Oshamanbe, Hokkaido, Japan, as this material contains large amounts of molecular H_2O (Fukuda and Nakashima 2008; Fukuda et al. 2009a, b). The IR spectra were measured at room temperature using a Fourier transform IR microspectrometer (JASCO IRT-3000) equipped with an autosample stage (IPS-3000) and a mercury–cadmium–telluride detector. Unpolarized IR light comes through the main spectrometer (FT/IR-6100) equipped with a silicon–carbide IR source and CaF_2 beamsplitter. The aperture size was either 30×30 μm or 50×50 μm, and the same step distances were applied during the mapping measurements. The spectral resolution was 4 cm^{-1}, and each spectrum was averaged from 128 scans. The thicknesses of the XZ section samples, measured with a micrometer, were in the range $80–120 \pm 5$ μm, which was suitable for both textural observations and good IR signals. Before the IR measurements, the samples were heated to ~120 °C for ~1 h to remove molecular H_2O from open pores. The mapping measurements were taken on areas that included quartz and phyllosilicates previously observed under a polarizing microscope and then a microscope equipped with the IR microspectrometer. These visible minerals can therefore be taken as marker locations to help interpret the spectral contrasts in the mapping results. The IR spectra were normalized to 1-cm sample thicknesses. Integral absorbances and band

heights of water stretching bands in the range of 2850–3750 cm^{-1} were determined using linear baselines in the same region. The linear baseline correction is based on an assumption that scattering and absorption of IR linearly depend on wave number by many possible factors such as sample color, grain size, and the focus and refraction of the IR passing through the optical system: For example, when light enters the sample, a wave of shorter wavelength scatters more at the surface of the sample than a wave of longer wavelength. It then undergoes wavelength-dependent refraction as it passes through the sample, and then leaves the sample. As a consequence, the absorbance increases linearly with progressively shorter wavelengths (i.e., higher wave numbers).

Differences among the IR calibrations

We compare six previously reported IR calibrations for determining water contents. In IR spectra, the OH fundamental stretching vibrations, seen at wave numbers of 2800–3750 cm^{-1}, are often used, and combination bands and overtone bands are also sometimes used. The Beer–Lambert law can then be used to convert absorbance to concentration, as follows:

$$A_{\mathrm{h}} = \varepsilon_{\mathrm{h}} c t \tag{1}$$

where A_{h} is the band height at a specific wave number, ε_{h} is the absorption coefficient, c is the concentration, and t is the thickness. Instead of band height, integral absorbance is often used, as follows:

$$A_{\mathrm{int}} = \varepsilon_{\mathrm{int}} c t \tag{2}$$

where A_{int} is the integral absorbance and $\varepsilon_{\mathrm{int}}$ is the absorption coefficient.

Many calibrations have been proposed to determine the water content in quartz using IR spectroscopy (Kats 1962; Paterson 1982; Aines et al. 1984; Nakashima et al. 1995; Libowitzky and Rossman 1997; Stipp et al. 2006; Thomas et al. 2009), and they have been discussed recently in connection with studies concerning the crystallographic orientations of quartz and the orientations of irradiated polarized IR light (e.g., Stalder and Konzett 2012; Baron et al. 2015; Frigo et al. 2016). The calibrations of Paterson (1982) and Libowitzky and Rossman (1997) do not require information on the mineral species, and they assume that the absorption coefficients increase linearly with wave number. The other calibrations listed above were determined specifically for quartz. Each proposed calibration uses different units such as absorption coefficients of L/mol H cm^2 or L/mol H_2O cm^2 ppm where the former is converted to the latter by multiplying by two, or simplified equations with factors that include specific sample thickness, quartz density, etc. The calculated water contents are also given variously as ppm H/Si or ppm H_2O. It is difficult, therefore, to see how the use of such a varied mix of calibration methods can provide reliable and consistent values of water contents. The differences in the results that arise from the use of different calibrations have seldom been discussed or reported in the literature, and each author has seemingly chosen a particular calibration on an ad hoc basis. However, exceptions are Stipp et al. (2006) and Kilian et al. (2016), who compared the water contents calculated with different calibrations.

In this paper, we express the water contents in the quartz as wt ppm H_2O, and the obtained values of integral absorbance and absorption coefficients, given as L/mol H_2O cm^2, are converted to wt ppm H_2O as follows:

$$\text{wt ppm } H_2O = \frac{M_{H_2O} A_{\mathrm{int}}}{t \rho \varepsilon_{\mathrm{int}}} \times 10^6 \tag{3}$$

where M_{H_2O} is the molar mass of H_2O (18 g/mol), A_{int} is the integral absorbance, t is the sample thickness in centimeters, ρ is the density of quartz (2650 g/L), and $\varepsilon_{\mathrm{int}}$ is in L/mol H_2O cm^2. By using M_{H_2O} and the molar mass of SiO_2, 1 wt ppm H_2O converts to 6.67 ppm H/Si. The calibrations we compare are given in Table 1 in terms of the same unified unit, which is an integral absorption coefficient (L/mol H_2O cm^2). Some of the original calibrations are given in different units and/or expressions. We describe below the procedures we used for making comparisons and give a unified unit.

The calibrations of Kats (1962) and Aines et al. (1984) use factors related to integral absorbance to calculate the water content in ppm H/Si. Using Eq. (3), their calibrations gave the equivalent absorption coefficients of 55,800 and 43,100 L/mol H_2O cm^2, respectively (Table 1).

Nakashima et al. (1995) measured the IR spectra of water in Brazilian agate and in the recrystallized quartz grains of the Sanbagawa metamorphic rocks from the same locations we sampled along the Asemi River. They showed that the water stretching bands could be attributed mainly to molecular H_2O. In their water content calculations, they used maximum band heights around 3400 cm^{-1}, the absorption coefficient of 81 L/mol cm for liquid water given by Thompson (1965), M_{H_2O}, and the density of quartz. As a result, they obtained the calibration shown by Eq. (1) (Table 1). To compare the calibration of Nakashima et al. (1995), which uses band heights, with the other calibrations that use integral absorbances (Kats 1962; Paterson 1982; Aines et al. 1984; Libowitzky and Rossman 1997; Stipp et al. 2006; Thomas et al. 2009), we quantitatively related integral absorbances to band heights in the regions of the water fundamental stretching vibrations. The IR spectra in Fig. 3a show different integral absorbances at 2800–3750 cm^{-1} and band heights around 3400 cm^{-1} for quartz and feldspar from

Table 1 List of proposed calibrations

Study	Equation	ε_{int} [L/mol H_2O cm^2]	Ratio of *c*
Kats (1962)	$c = 0.812 \cdot A_{int}/t$ [ppm H/Si]	55,800	0.68
Aines et al. (1984)	$c = 1.05 \cdot A_{int}/t$ [ppm H/Si]	43,100	0.88
Nakashima et al. (1995)	$c = 84 \cdot A_h/t$ [wt ppm]	33,200[a]	1.15
Paterson (1982)	$\varepsilon_{int\ eff} = A_{int}/c \cdot t$ [L/mol H_2O cm^2]	38,000[b]	1.00
Paterson (1982) simplified[c]	$\varepsilon_{int} = 150 \cdot \gamma \cdot (3780 - \nu_{mean})$ [L/mol H cm^2]	38,000 ($\nu_{mean} = 3400$, $\gamma = 1/3$)	1.00
Libowitzky and Rossman (1997)	$\varepsilon_{int} = 246.6 \cdot \gamma \cdot (3753 - \nu_{mean})$ [L/mol H_2O cm^2]	29,000 ($\nu_{mean} = 3400$, $\gamma = 1/3$)	1.31
Stipp et al. (2006)	$c = 100 \cdot M \cdot A_{int}/(t \cdot \rho \cdot \varepsilon_{int})$ [wt% H_2O]	24,100	1.58
Thomas et al. (2009)	$c = A_{int}/\varepsilon_{int} \cdot t$ [wt% H_2O]	89,000	0.43

The equations given are the originals from each paper. The unit referred to on the left side of each equation is given in brackets

[a] The absorption coefficient of Nakashima et al. (1995) using band heights was converted to a coefficient for integral absorbance using Eq. (4)

[b] The water content *c* was determined using Eq. (5)

[c] This follows the method of Libowitzky and Rossman (1997) and was used for the IR mapping data. See the text for further details

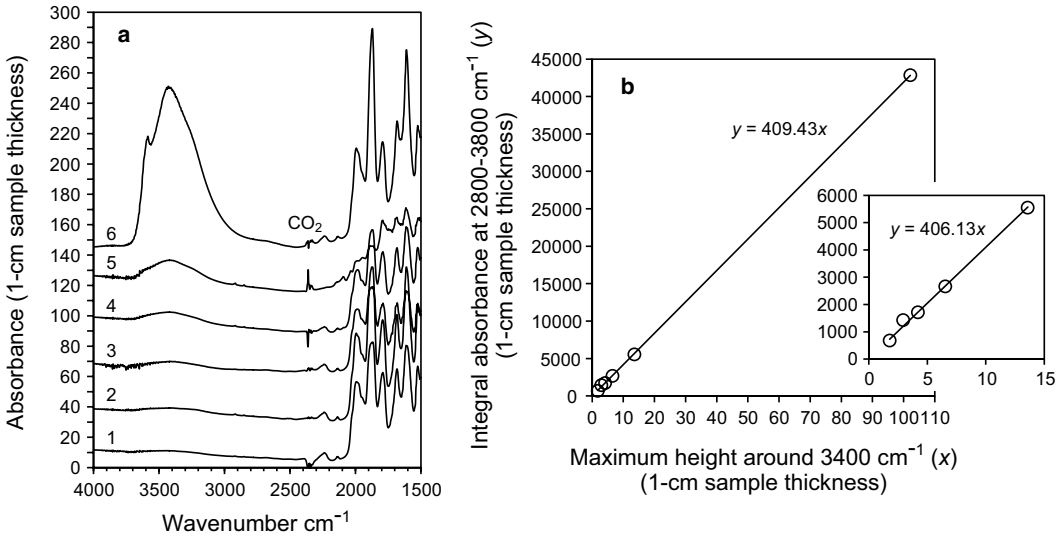


Fig. 3 **a** IR spectra showing different absorbances of water bands at 2800–3750 cm^{-1}. Spectra numbers: 1–4 = quartz from the Asemi River, with 1 = sample No. 7-35, 2 = No. 7-49, 3 = No. 7-89, and 4 = No. 7-35 (healed cracks; see the IR mapping result in Fig. 8 and the text); 5 = albite porphyroblast with minor quartz (No. 7-35); 6 = chalcedony from Oshamanbe. The structural vibrations < 2300 cm^{-1} roughly distinguish mineral species. The spectra show typical stretching bands of molecular H_2O. Atmospheric CO_2 bands are seen around 2360 cm^{-1}. The IR spectrum of chalcedony has a weak Si–OH band at 3585 cm^{-1}. **b** Relationship between band heights and integral absorbance of molecular H_2O. The left figure includes data for chalcedony that show much higher values than the others. The right figure shows only data for the Asemi River samples

the Asemi River and chalcedony. The spectra commonly show broad bands due to molecular H_2O. The IR spectra for the chalcedony show a much higher absorbance than in the other samples. They also include a weak Si–OH band at 3585 cm^{-1}, which is typical of chalcedony and agate and associated with dislocation Si–OH and surface Si–OH at grain boundaries because of fine grains (e.g., a few hundred nanometers) (e.g., Flörke et al. 1982; Frondel 1982; Graetsch et al. 1985; Nakashima et al. 1995). The contribution of this band to the integral absorbance is not significant. The maximum band heights around 3400 cm^{-1} and the integral absorbances at 2800–3750 cm^{-1} show a good correlation (Fig. 3b), regardless of whether the data include or exclude chalcedony. The trend shown in Fig. 3b is valid, since the feature of molecular H_2O should be preserved for any materials that show clear molecular H_2O bands. According to the slope in Fig. 3b, the relationship between band height and integral absorbance can be simplified as follows:

$$A_{\text{int at 2800–3750 cm}^{-1}} = 410 \times A_{\text{h at }\sim\text{3400 cm}^{-1}} \quad (4)$$

Using this relationship and Eq. (3), the integral absorption coefficient converted from the original calibration of Nakashima et al. (1995) is 33,200 L/mol H_2O cm^2 (Table 1).

The calibration of Paterson (1982) is given as follows:

$$c = \int \frac{A_{\mathrm{h\ at}\ \upsilon}/t}{150 \cdot \gamma \cdot (3780 - \nu)} \mathrm{d}\nu \tag{5}$$

where $A_{\mathrm{h\ at}\ \nu}$ is the absorbance at the wave number ν (cm^{-1}), t is the thickness in cm, and γ is the orientation factor. The value of $\gamma = 1/3$ is used for H_2O, as suggested by Paterson (1982), assuming an isotropic OH distribution, any beam direction, and any polarization. The denominator refers to the absorption coefficient, which depends linearly on the wave number, so that $\varepsilon_{\mathrm{h}} = 150 \cdot \gamma \cdot (3780 - \nu)$ [L/mol H cm^2]. Paterson (1982) also introduced the effective integral absorption coefficient as follows:

$$\varepsilon_{\mathrm{int\ eff}} = \frac{A_{\mathrm{int}}}{ct}, \tag{6}$$

where c is from Eq. (5) and A_{int} is given as follows.

$$A_{\mathrm{int}} = \int A_{\mathrm{h\ at}\ \upsilon} \mathrm{d}\nu. \tag{7}$$

In the calibration of Libowitzky and Rossman (1997), the mean wave number (ν_{mean}) was applied, and this is given by band deconvolution of the water stretching bands with integral absorbance of each band at each wave number. The IR spectra of our quartz samples show broad H_2O stretching bands at 2800–3750 cm^{-1} with the maximum around 3400 cm^{-1} (Fig. 3), which can also be the mean. We used 3400 cm^{-1} as the ν_{mean}. Then, the calibrations become $\varepsilon_{\mathrm{int}} = 246.6 \cdot \gamma \cdot (3753 - \nu_{\mathrm{mean}})$ in L/mol H_2O cm^2 using Libowitzky and Rossman (1997) (γ is introduced in our study but was not employed in the original paper), or $\varepsilon_{\mathrm{int}} = 300 \cdot \gamma \cdot (3780 - \nu_{\mathrm{mean}})$ using Paterson (1982), thus giving 29,000 and 38,000 L/mol H_2O cm^2, respectively (Table 1). The differences from the original calibration of Paterson (1982) are $\pm \sim 10\%$ in $\varepsilon_{\mathrm{int\ eff}}$ as given by Eq. (6), and consequently water content, as given by Eq. (5). This simplification of the calibration of Paterson (1982) is helpful for comparing all the calibrations and for calculating water contents from large data sets of IR mapping, as we will show later.

Details of using all the various calibrations are given in Table 1, where the ratio of c for each calibration means the ratio of $1/\varepsilon_{\mathrm{int}}$ from Eq. (2) (and Eqs. 1 and 4 for Nakashima et al. 1995). The ratio of c in all calibrations varies from 0.43 (Thomas et al. 2009) to 1.58 (Stipp et al. 2006). This is also shown in Fig. 4 in the form of the determined integral absorbance. The large

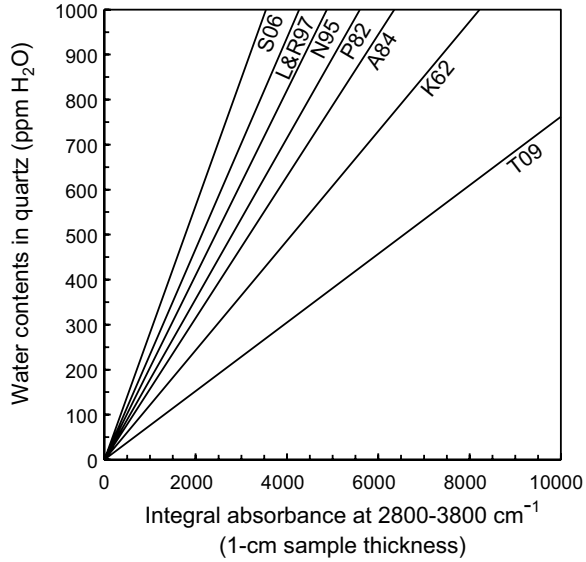


Fig. 4 Different calibrations used to calculate water contents and the integral absorbance of molecular H_2O. See Table 1 for the calibrations

differences among the calibrations may be associated roughly with the differences in the water species. Thomas et al. (2009) reported that the IR spectra for quartz consist mainly of OH sharp bands and that the value of $\varepsilon_{\mathrm{int}}$ is 89,000 L/mol H_2O cm^2 (the maximum in Table 1), thus giving the lowest calculated water contents. Kats (1962) also showed OH sharp bands and an $\varepsilon_{\mathrm{int}}$ value of 55,800 L/mol H_2O cm^2. Aines et al. (1984) reported a higher contribution of molecular H_2O for synthetic quartz samples, with an $\varepsilon_{\mathrm{int}}$ value of 43,100 L/mol H_2O cm^2. The calibration of Paterson (1982) was designed for molecular H_2O and assumes the orientation factor. The value of $\varepsilon_{\mathrm{int}}$ is 38,000 L/mol H_2O cm^2. The calibration of Nakashima et al. (1995) is based on the ε_{h} given by Thompson (1965) just for molecular H_2O, and the value of $\varepsilon_{\mathrm{int}}$ is 33,200 L/mol H_2O cm^2 using Eq. (4). These two calibrations therefore yield similar $\varepsilon_{\mathrm{int}}$ values. Broad bands were also reported by Stipp et al. (2006) with several –OH sharp bands ($\varepsilon_{\mathrm{int}}$ of 24,100 L/mol H_2O cm^2). Thomas et al.'s (2009) $\varepsilon_{\mathrm{int}}$ value of 89,000 L/mol H_2O cm^2 was determined for crystallographically oriented samples and considering the orientations of –OH in the quartz crystal structure. As carefully discussed in recent papers (Stalder and Konzett 2012; Baron et al. 2015; Frigo et al. 2016), the values of $\varepsilon_{\mathrm{int}}$ for randomly distributed molecular H_2O given by Paterson (1982), Libowitzky and Rossman (1997), and Stipp et al. (2006) are for unpolarized IR light irradiated to the sample. Therefore, Paterson (1982) (and for the data of Libowitzky and Rossman 1997 in the present study) applied an

orientation factor $\gamma = 1/3$, assuming an isotropic distribution of the water in quartz. Consequently, three times the ε_{int} values of Paterson (1982), Libowitzky and Rossman (1997), and Stipp et al. (2006) ($38{,}000 \times 3 = 114{,}000$, $29{,}000 \times 3 = 87{,}000$, and $24{,}100 \times 3 = 72{,}300$ L/mol H_2O cm^2, respectively) can give rather similar values to the ε_{int} value given by Thomas et al. (2009) (89,000 L/mol H_2O cm^2). The same argument can be applied to the ε_{int} value given by Thompson (1965), which with our modification is $33{,}200 \times 3 = 99{,}600$ L/mol H_2O cm^2. In Aines et al. (1984), their ε_{int} includes one direction twice in their determination of the ε_{int}. Thus, twice their ε_{int} value ($43{,}100 \times 2 = 86{,}200$ L/mol H_2O cm^2) gives a similar value to that of Thomas et al. (2009). Please also see the detailed discussions in Stalder and Konzett (2012), Baron et al. (2015), and Frigo et al. (2016). In summary, molecular H_2O may give lower values of ε_{int} than the –OH species when there are specific orientations of the water species in the quartz crystal structure relative to the orientations of the irradiated IR light. Thus, when IR spectra are used to determine the water content, one first needs to clarify the spectral features with contributions of molecular H_2O and the –OH species, and at present we also need to show how each calibration gives different values of water content depending on the values of the absorption coefficients.

Determination of water content distribution from the IR mapping results

To calculate the water contents from the large data sets of IR mapping, we used the simplified calibration of Paterson (1982) following the approach of Libowitzky and Rossman (1997). We used the ν_{mean} value of 3400 cm^{-1} and the orientation factor of $\gamma = 1/3$ designed for H_2O (Table 1), because the IR spectra of the quartz in our samples show dominant H_2O bands (Fig. 3). As mentioned in the previous section, this method is for unpolarized IR light plus randomly orientated water. We note that the values of ν_{mean} and the orientation factor would change if other –OH components in the IR spectra were introduced (e.g., contributions from phyllosilicates in our specimens). However, in order to focus on the water contents in quartz in our mapping results, Eq. (3) for quartz was applied. Therefore, the calculated water contents for aggregates including quartz and some other minerals give rough or tentative estimates. Some spectra include weak organic bands around 2900 cm^{-1} that would be due to the resin used for polishing, but these bands do not affect the water bands.

Results

We performed IR mapping of quartz-rich regions in specimens of the Sanbagawa metamorphic rocks with different grades of metamorphism. Figure 5 shows the results for the sample from the chlorite zone (No. 7-93), the lowest grade metamorphic rock. The measured area in Fig. 5 includes phyllosilicates (Fig. 5a, b). The quartz

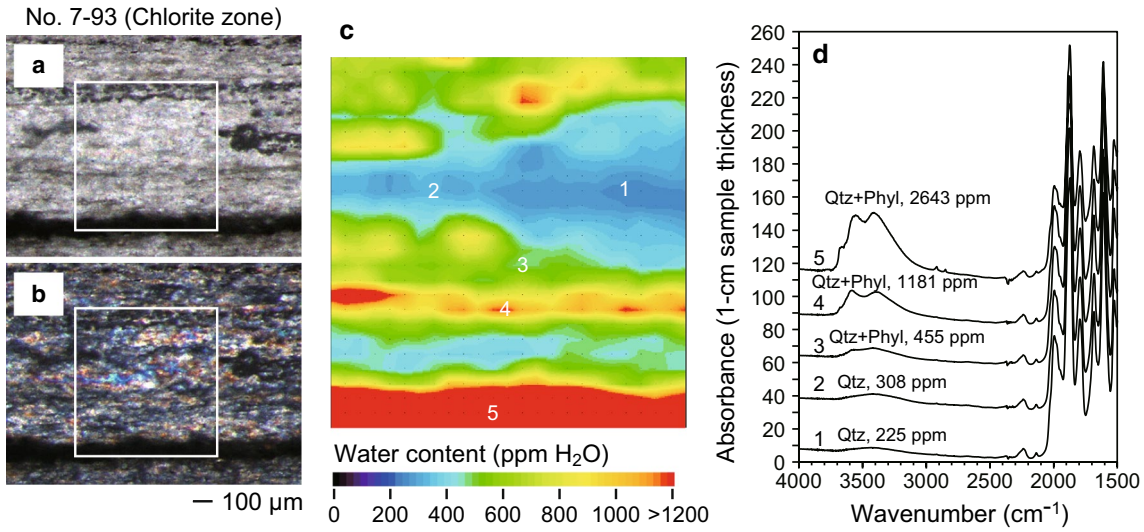


Fig. 5 IR mapping result for chlorite-zone specimen No. 7-93. **a** Optical photomicrograph, plane-polarized light. **b** Cross-polarized light. The IR mapped areas are shown by squares. **c** Water distributions using the calibration of Paterson (1982). Numbers represent raw IR spectra in **d**. Spectra for the lowest (line 1) and highest (line 2) water contents are shown. Other spectra are mixtures of quartz (Qtz) and phyllosilicates (Phyl) and show large variations in band components and water contents

grain size is < 40 μm, the sample thickness ~ 100 μm, and the aperture size for IR spectroscopy 30 × 30 μm. Therefore, the IR-measured area includes quartz grain boundaries where molecular H_2O might be trapped. The water distribution according to the IR mapping measurements shows that when only the quartz aggregate is analyzed, the water contents range from 225 to 308 wt ppm H_2O with the representative value of ~ 280 ppm (namely, the value that is widely seen in the measured area) (Fig. 5c, d). When phyllosilicates are included, the water contents become much higher (> 400 ppm, as in spectra 3–5 in Fig. 5d). The spectra of water bands in these situations consist of bands around 3560 and 3640 cm^{-1} on a broad band of molecular H_2O. All these bands would be associated with chlorite, and the nature of the components and their wave numbers probably change according to the type of phyllosilicate and its composition, among other factors (e.g., Oinuma and Hayashi 1965; Prieto et al. 1991). The structural vibrations of phyllosilicates at < 2300 cm^{-1} are not clearly seen because of the dominant structural bands of quartz (Fig. 5). However, when large amounts of phyllosilicate are included (showing high water stretching absorption bands), the component of the quartz structural bands decreases slightly, especially around 1990 cm^{-1}. Phyllosilicates are commonly observed in the optical photomicrographs, and this is reflected in the corresponding IR mapping results. However, some phyllosilicates may appear dusty or be obscured in an optical photomicrograph (e.g., be located in a deep part of the thin section and/or the contents are low), but this problem is addressed by paying attention to other well-observed and identified phyllosilicates in the area. Even when a phyllosilicate grain is not clearly defined, the relatively broad bands around 3560 and 3640 cm^{-1}, which may be combined, often reveal its presence (e.g., Spectrum 3 in Fig. 5c, d).

The mean grain size of quartz in the garnet zone sample (No. 7-45) is ~ 80 μm (Fig. 6b), which is ~ 40 μm larger than in the chlorite zone. The range of water contents in the quartz is 94–210 ppm with the representative value of ~ 190 ppm (Fig. 6c, d), which is 90 ppm lower than in chlorite-zone specimen No. 7-93. The IR spectra exhibit a band at 3610 cm^{-1} in the measured area, when phyllosilicates are included (lines 3–5 in Fig. 6d). With increasing amounts of phyllosilicate, the broad band at 2800–3750 cm^{-1}, which would probably include both H_2O and other –OH components, becomes high. The spectrum around 3400 cm^{-1} appears to be more intense than the typical H_2O band and has shoulders around 3425 and 3020 cm^{-1}. Similar spectral features were reported by Kilian et al. (2016). When these bands due to phyllosilicates (most likely biotite in this case) become high, the component of the structural vibrations of quartz, especially around 1990 cm^{-1}, becomes less than other bands, probably because of the structural vibrations of the phyllosilicate.

In the oligoclase–biotite schist (specimen No. 7-26), which represents the highest metamorphic grade in the study area, the measured area includes a single host grain of quartz (> 500 μm) with recrystallized grains of < 120 μm (Fig. 7). Water contents of 39–87 ppm with

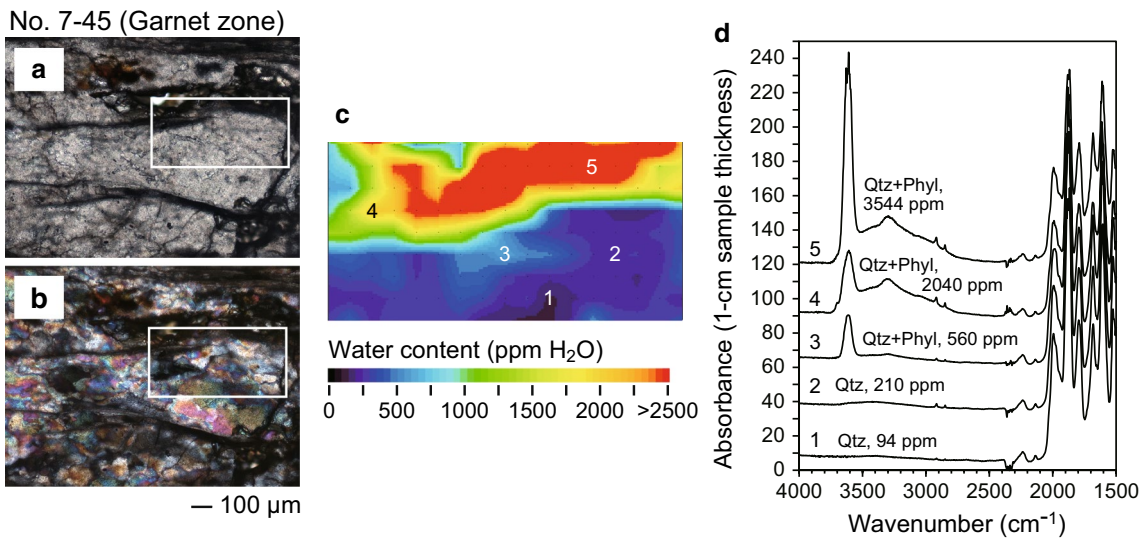


Fig. 6 IR mapping result for garnet zone specimen No. 7-45. **a** Optical photomicrograph, plane-polarized light. **b** Cross-polarized light. The IR mapped areas are shown by squares. **c** Water distributions using the calibration of Paterson (1982). Numbers represent raw IR spectra in **d**

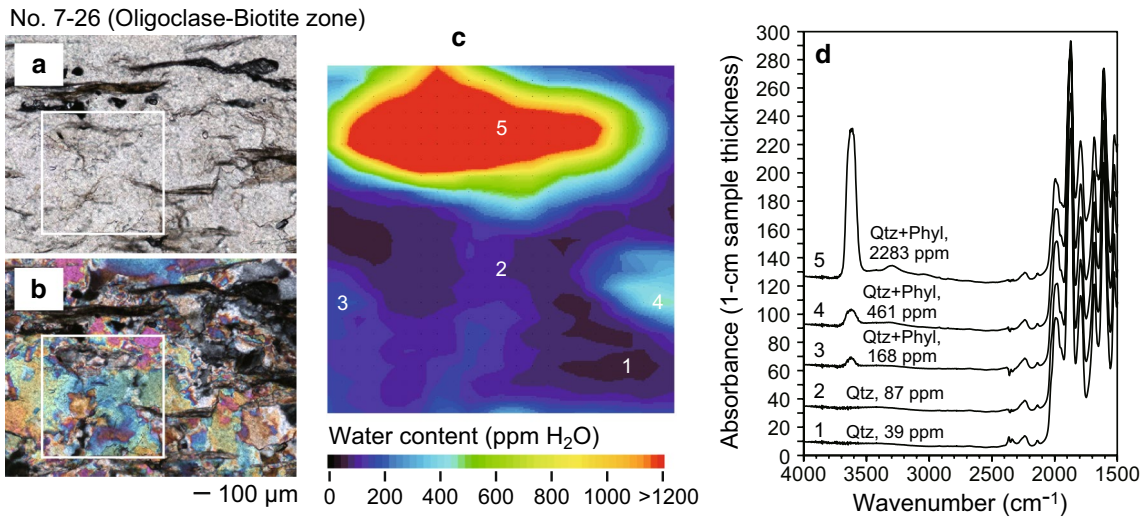


Fig. 7 IR mapping result for oligoclase–biotite zone specimen No. 7-26, which represents the highest grade of metamorphism in the study area. **a** Optical photomicrograph, plane-polarized light. **b** Cross-polarized light. The IR mapped areas are shown by squares. **c** Water distributions using the calibration of Paterson (1982). Numbers represent raw IR spectra in **d**

the representative value of ~70 ppm were measured, the lowest of all the samples (Fig. 7c, d). The relationship between the water contents in the host quartz grain and those in the recrystallized grains is unclear.

Sample No. 7-35 from the albite–biotite zone includes quartz grains that are > 120 µm in size (Fig. 8a, b). The sample shows alignments of fluid inclusions (Fig. 8a) that are not associated with features of plastic deformation such as undulose extinction or subgrains, and the fluid inclusions would have been trapped after the plastic deformation took place, during a stage of brittle deformation that was accompanied by an influx of water and crack healing. The lowest water content is ~40 ppm, which is similar to that in the highest metamorphic grade sample (No. 7-26, Fig. 7). The areas with fluid inclusions show higher water contents of up

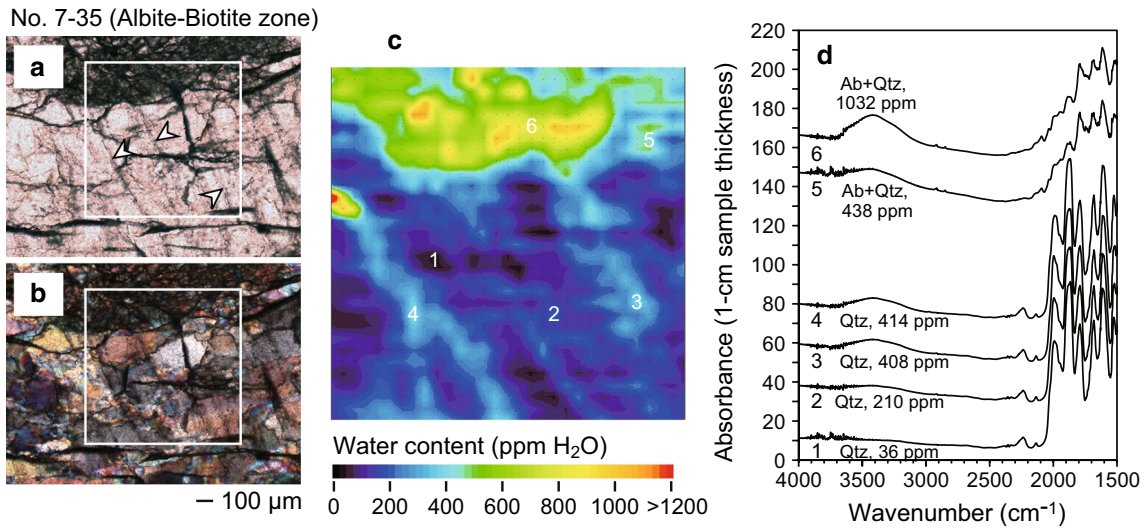


Fig. 8 IR mapping result for albite–biotite zone specimen No. 7-35, which shows aligned fluid inclusions. **a** Optical photomicrograph, plane-polarized light. **b** Cross-polarized light. Cracks are also seen. The aligned fluid inclusions in the quartz are indicated by arrows. The IR mapped areas are shown by squares. **c** Water distributions using the calibration of Paterson (1982). Numbers represent raw IR spectra in **d**. Aligned fluid inclusions show relatively high values (lines 3 and 4)

to ~410 ppm and exhibit the same spectral features of molecular H_2O. The water contents of an albite porphyroblast (with minor quartz) are 400–1100 ppm (Fig. 8c), and the IR spectra show typical molecular H_2O bands (Fig. 8d).

Discussion

The relationship between water content of quartz and metamorphic grade is shown in Fig. 9. The distances in kilometers are to the north from the intersection of the Yoshino and Asemi Rivers (Fig. 1), and each location was traced on a straight line. The data for specimen No. 7-35 are not included since the measured area for that specimen included complicated alignments of secondary fluid inclusions in healed cracks (Fig. 8). The water contents measured by Nakashima et al. (1995) are also shown, after modifying their calibration to fit with the calibration of Paterson (1982) (see Eq. (4) and Table 1).

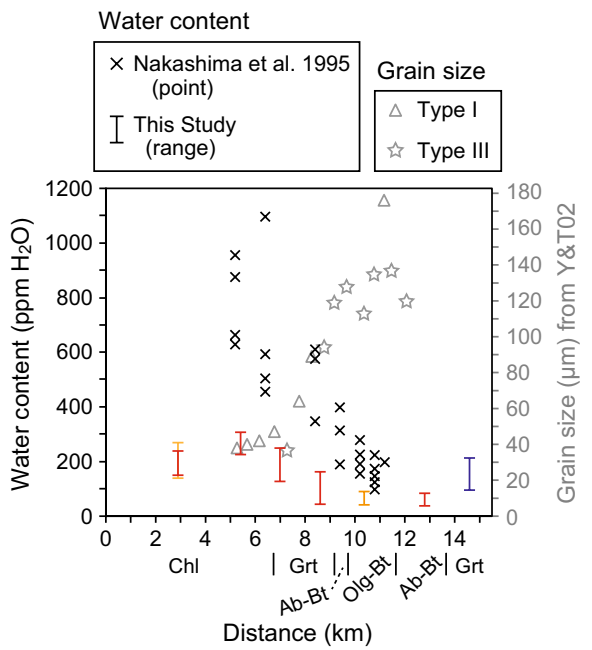


Fig. 9 Water distributions in quartz in Sanbagawa metamorphic rocks of various metamorphic grades along the Asemi River. The ranges of water content obtained by us using IR mapping are shown. Data from Nakashima et al. (1995) obtained by point analyses are also shown after modifying their calibration to fit the calibration of Paterson (1982). Data for the grain sizes of quartz and grain morphologies are from Yagi and Takeshita (2002). (They reported type-I and type-III grains, but no type-II grains.) Note that the locations and textures of their specimens are not exactly the same as ours. The distances in kilometers are measured to the north of the intersection between the Yoshino and Asemi Rivers (Fig. 1), and each location was traced along a straight line. For our samples, blue indicates pelitic rock, yellow indicates psammitic rock, and red indicates siliceous rock

Data for the average grain size of quartz and microstructural types are from Yagi and Takeshita (2002). Our IR results show that the water contents of quartz are highest at ~310 ppm H_2O in the chlorite zone, which represents the lowest metamorphic grade. The water contents gradually decrease with increasing metamorphic grade, and they are lowest at ~40 ppm in the oligoclase–biotite zone. In the same area, Nakashima et al. (1995) also showed that the water contents of quartz decrease with increasing metamorphic grade, but the values of water content they determined are much higher than ours. Again, we modified their calibration to fit the calibration of Paterson (1982) used in the present study, and we directly compared their water content values with ours, as shown in Fig. 9. The water contents they determined using point analyses, and adjusted by us, decrease from 450–1100 ppm in the chlorite zone to 90–240 ppm in the biotite zone. They argued that because lower-grade rocks have smaller grain sizes, the grain boundary volumes are correspondingly large, and that it is along the grain boundaries where larger amounts of molecular H_2O can be trapped. However, numerous grains of phyllosilicates (chlorite and mica) are spaced irregularly between the quartz grains in the low-grade samples, and only limited areas could be chosen for the IR analyses, but even so, the IR mapping inevitably included tiny phyllosilicate grains, which have higher water contents than the quartz (Figs. 5, 6, 7, 8). The IR point measurements of Nakashima et al. (1995), especially for samples of low-grade metamorphic rocks, may therefore have included phyllosilicates as well as small quartz grains.

Ito and Nakashima (2002) similarly performed IR mapping analyses of natural chert and shale in the Northern Chichibu Belt of the Kanto Mountains, Central Japan, as well as natural Brazilian agate, and they derived a relationship between water content and grain size, and estimated grain boundary width filled with molecular H_2O, which we will discuss later. The Northern Chichibu Belt is the lowest grade part of the 'Sanbagawa Metamorphic Belt' in its broad sense (Shimizu 1988). Some of their IR spectra show a band around 3600 cm^{-1}. In view of the above discussion, we suggest this band as well as the molecular H_2O band may be attributed to phyllosilicates. The band around 3600 cm^{-1} can also be attributed to dislocation Si–OH that has been found in dynamically recrystallized natural quartz (Niimi et al. 1999; Gleason and DeSisto 2008) and experimentally deformed quartz single crystals (Stünitz et al. 2017). Stünitz et al. (2017) showed that this band disappears after annealing due to the disappearance of dislocations. The band related to dislocation Si–OH is much sharper than the band due to –OH in phyllosilicate (Figs. 5, 6, 7). Contents of dislocation Si–OH in quartz under various pressure and

temperature conditions are estimated from thermodynamics to be less than 100 H/10^6Si (i.e., 15 wt ppm H_2O), assuming Si–O bonds are replaced by Si–OH (namely, H_4O_4 tetrahedra) (Paterson 1986). In contrast, IR studies of chalcedony and agate have estimated H_2O and Si–OH contents of >1 wt% each (Flörke et al. 1982; Graetsch et al. 1985). These values were estimated from the clear stretching and bending combination bands around 5200 and 4500 cm^{-1} in their IR spectra. The H_2O and Si–OH contents of our chalcedony were estimated to be ~0.3 wt% each (Fukuda et al. 2009a). In addition to the dislocation Si–OH in chalcedony, Si–OH can also be associated with surface Si–OH at grain boundaries because of the fine grain size (e.g., a few hundred nanometers). In any case, previous IR studies of quartz demonstrated that even when only dislocation Si–OH was included, the band at 3585 cm^{-1} was observed (Kronenberg and Wolf 1990; Kronenberg et al. 1990; Kronenberg 1994; Niimi et al. 1999; Gleason and DeSisto 2008; Stünitz et al. 2017). However, when phyllosilicates are not included in the analyses of our schist samples, the band due to Si–OH is not seen. This may indicate that the dislocation density in the quartz schists was lowered by dynamic recovery, and that the influence of dislocation Si–OH was not important in the part of the Sanbagawa Metamorphic Belt examined in the present study.

Finch et al. (2016) demonstrated that the water content of quartz decreases with the development of mylonite and ultimately ultramylonite, where the mean matrix grain size decreases from 95 to 60 μm as a result of dynamic recrystallization. They measured only intragranular water, which means that no grain boundaries were included. Their IR spectra for quartz also show the dominant stretching vibration bands of molecular H_2O. The water contents they reported using the calibration of Kats (1962) ranged from 3265 ppm H/Si in weakly deformed quartz to 1455 ppm H/Si in the ultramylonite. These values correspond, respectively, to ~720 and 320 ppm H_2O, using the calibration of Paterson (1982) (Table 1). The absolute values could be related to the water environment of the original quartz grains and the conditions of metamorphism and deformation, so it is not possible to directly compare them with our results. The most interesting finding of Finch et al. (2016) was that the intragranular water content decreases as dynamic recrystallization progresses, and they proposed that the water was swept out during the recrystallization process (subgrain rotation or grain boundary migration, depending on the sample). On the other hand, our higher-grade samples taken from the garnet and biotite zones show lower water contents down to ~40 ppm H_2O (Fig. 9), which would be associated with the minimum intragranular water contents of the quartz grains. It is possible that the reduction in water content with increasing metamorphic grade was linked to the sweeping out of water with progressive metamorphism. In the chlorite zone, where the mean quartz grain size is ~40 μm, the water contents are up to 310 ppm, which would be largely due to water along grain boundaries as discussed below.

Microstructures of quartz schists along the Asemi River show large variations under the microscope, consistent with descriptions made previously (Masuda 1982; Hara et al. 1988, 1992; Yagi and Takeshita 2002). Laboratory studies on quartz aggregates have revealed that the size and morphology of recrystallized grains depend on temperature and strain rate (Masuda and Fujimura 1981; Hirth and Tullis 1992). The dislocation creep regimes 1 to 3 of Hirth and Tullis (1992) describe the processes of grain size reduction in coarse-grained quartzite samples, whereas the P- and S-types of Masuda and Fujimura (1981) represent steady-state microstructures resulting from overall coarsening of agate samples. The relationships between these different classifications are explained in Fig. 5 of Shimizu (2008). Quartz schists in the Sanbagawa Metamorphic Belt are metacherts, and therefore regimes 1 to 3 of Hirth and Tullis (1992) are not applicable. However, the development of subgrains and the occurrence of lobate grain boundaries indicate that both subgrain rotation and migration recrystallization (that characterize regime 3) were active in all the quartz schists studied. Experimental and theoretical studies of dynamic recrystallization indicate a reduction in average grain size with increasing strain rate and/or decreasing temperature, both of which result in increasing differential stress (e.g., Shimizu 2008, 2011, 2012). The relationships between average grain size and metamorphic grade are, at a first approximation, understood by the temperature dependence of the dynamically recrystallized grain size. The amount of grain boundary water would be higher in rocks of lower metamorphic grade.

Assuming cubic or tetradecahedral grain shapes, the water content w at a grain boundary in wt ppm is expressed as

$$w = 1132\,(\text{for a cube}) \text{ or } 894\,(\text{for a tetradecahedron}) \times \delta/d \tag{8}$$

where δ is the grain boundary width in nm and d is the grain size in μm (Ito and Nakashima 2002). Using this equation, the water contents in the chert and agate samples studied by Ito and Nakashima (2002) vary from ~300 wt ppm H_2O (for chert with the grain size of 17.5 μm) to ~9000 wt ppm H_2O (for agate with the grain size of 1 μm) (Fig. 8 of Ito and Nakashima 2002), following the calibration of Nakashima et al. (1995), which can be converted into ~260 wt ppm for the chert and ~7830 wt ppm for the agate using the calibration of

Paterson (1982) (Table 1). In the quartz schists of the present study, the lowest water content of 40 ppm H_2O was from a single quartz crystal that was visible under the optical microscope and larger than the aperture size of 30 or 50 μm of the IR microspectrometry equipment and larger than the sample thickness of 80–120 μm (Figs. 6, 7, 8). In contrast, the measured water contents of up to 310 ppm from the chlorite zone would have included both intragranular and grain boundary water. Assuming that the intracrystalline water contents are almost constant and that the observed differences in water contents in the quartz-rich parts are due to grain boundary water, the differences in the water contents from 310 ppm (intragranular and grain boundary water in the chlorite zone) to 40 ppm (intragranular water in the biotite zones) (Fig. 9) show that the grain boundary width was ~10 nm in all our samples. This value of grain boundary width is consistent with that estimated by Ito and Nakashima (2002) for the chert and agate, each of which has different water contents and grain sizes. However, it should be noted that some of their measurements with the relatively broad band around 3600 cm^{-1} may include phyllosilicates that would give high H_2O contents, as discussed above.

The amount of grain boundary fluid in the chlorite zone along the Asemi River was also estimated by Hiraga et al. (2001) on the basis of morphological observations of grain boundaries using high-resolution transmission electron microscopy (HTEM). They observed lens-shaped fluid at grain boundaries as well as fluid at triple junctions of combinations of quartz and/or albite grains (e.g., quartz–quartz–albite or quartz–albite–albite). Their estimation of the fluid volume from morphology was ~0.05 vol%. They also measured IR spectra and showed dominant H_2O bands similar to ours. Their water content calculations using the calibration of Nakashima et al. (1995) were consistent with those estimated from grain boundary morphology. Assuming that the grain boundary fluid was previously distributed uniformly, the average width of a grain boundary was estimated to be ~6 nm (Hiraga et al. 2012), which is comparable with our estimation based on IR measurements and an assumed grain morphology (Eq. 8). However, unlike polygonal structures that result from static grain growth (e.g., Fukuda et al. 2019), the actual grain shapes in dynamically recrystallized quartz rocks of the type-I and type-III are not close to cubes or tetradecahedra, and grain boundaries show curved or serrated morphologies (Takahashi et al. 1998). Moreover, grain size distribution in dynamic recrystallization is close to lognormal, which means that the grains smaller than the average size are large in number (Shimizu 1998, 1999). Hence, the grain boundary width evaluated by the above simple calculation is considered to be an upper limit. The width of a grain boundary without a fluid phase was estimated to be less than 0.5 nm by HTEM for granite ultramylonite from the Hatagawa Fault, NE Japan (Hiraga et al. 1999). This value gives the lower bound for grain boundary width. Since there are few data for the water contents of naturally deformed quartz, further studies are needed to understand the behavior of water in nature, as well as the textures that form under different conditions of deformation and metamorphism. However, our IR mapping results have shown that it is possible to make realistic correlations between the development of quartz textures and the contents of intragranular water in quartz or grain boundary water, while also considering the effects of phyllosilicate minerals. A knowledge of the differences in the various proposed IR calibrations, and their validity, is essential when estimating water contents in quartz.

Conclusions

We measured the water contents of deformed quartz in Sanbagawa metamorphic rocks of different metamorphic grade from the chlorite to biotite zones. We examined previously reported IR calibrations for determining the water contents of quartz, and we unified these previous calibrations as an absorption coefficient in L/mol H_2O cm^2. The value of this coefficient varies from 24,100 (Stipp et al. 2006) to 89,000 (Thomas et al. 2009), which roughly corresponds to the increase in the components of structural –OH in IR spectra (probably when –OH species are orientated in the quartz crystal structure) and to the decrease in components of molecular H_2O. (This could also be orientated in some cases.) We used the calibration of Paterson (1982) for molecular H_2O ($\varepsilon_{int} = 38{,}000$ L/mol H_2O cm^2). Our IR mapping results for quartz in the Sanbagawa metamorphic rocks, considering the progressive changes in texture, reveal that the water contents, dominantly molecular H_2O trapped in quartz aggregates, were as much as 310 ppm H_2O (2070 ppm H/Si) in the chlorite zone (including at grain boundaries) and as low as 40 ppm H_2O (270 ppm H/Si) in the biotite zone (possibly as intracrystalline water). We suggest that this progressive decrease in water content relates to the progressive decrease in the grain boundary volume as grain size increases. Thus, the maximum possible difference in water content (270 ppm H_2O) is the amount that can be trapped at grain boundaries, which gives an estimated grain boundary width of up to ~10 nm for a grain size of 40 μm.

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Authors' contributions
JF performed the analyses and drafted the manuscript with contributions from IS. IS collected samples and designed the framework of this study. Both authors read and approved the final manuscript.

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Availability of data and materials
The data are available from the corresponding author (JF) upon request (jfukuda@crystal.kobe-u.ac.jp). Samples and thin sections of quartz schists are held in the laboratory of the second author (IS) at Kyoto University.

Competing interests
The authors declare that they have no competing interests.

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