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Photochemical reaction playing a key role in particulate matter pollution over Central France: Insight from the aerosol optical properties

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Abstract

Atmospheric particle is one of the major air pollutants, and believed to be important for air quality, radiative forcing and climate. Measurements of aerosol optical properties, size distribution and PM\textsubscript{10} concentration were conducted at Orleans, central France during spring (7 March to 25 April) and autumn (25 October to 5 December) 2013. The average values of aerosol scattering coefficient ($b_{\text{sc}}$), absorption coefficient ($b_{\text{abs}}$), single scattering albedo (SSA) at 532 nm and PM\textsubscript{10} concentration are 54.9 ± 58.2 Mm\textsuperscript{-1}, 10.6 ± 10.9 Mm\textsuperscript{-1}, 0.81 ± 0.10 and 30.6 ± 21.6 μg/m\textsuperscript{3} for the spring campaign, and 35.4 ± 36.7 Mm\textsuperscript{-1}, 3.9 ± 4.4 Mm\textsuperscript{-1}, 0.83 ± 0.13 and 17.4 ± 11.8 μg/m\textsuperscript{3} for the autumn campaign, respectively. During the whole observation, the air parcel transported from Atlantic Ocean plays a role in cleaning up the ambient air in Orleans, while the air mass
coming from the Eastern Europe induces the pollution events in Orleans. In this study, a simple approach, which based on the diurnal variation of PM$_{10}$ concentration, Boundary layer depth (BLD) and the human activity factor derived from anthropogenic emission rate, was introduced to estimate the contribution of secondary aerosol to ambient aerosols. Our results show that secondary particles formation trigged by photochemical reactions and oxidations can contribute maximum of 64% and 32% for PM$_{10}$ mass concentration during the spring and autumn time, respectively. These results highlight that photochemical reactions can enhance the atmospheric oxidation capacity and may faster the secondary particle formation and then play an important role in air quality.

**Keywords:** Photochemical reactions, Optical properties, Size distribution, Orleans-France, Secondary formation.

**Introduction**

Atmospheric particles play an important role in affecting air quality, radiative forcing and climate (Charlson et al., 1992). Aerosol particles influence climate both directly, by scattering and absorbing solar and infrared radiation (Charlson et al., 1992; Haywood et al., 1997), and indirectly, by altering the albedo and lifetime of clouds (Kaufman et al., 2005). In addition, the visibility (Vis), as the most readily perceived
impact of air pollution, is highly dependent on the aerosol loading, and can be calculated by aerosol extinction coefficient \( (b_{ext}) \), i.e., \( \text{Vis}=3.912/b_{ext} \) (Seinfeld and Pandis, 2006).

The new particle formation (NPF) and secondary particle (including inorganic and organic) formation processes in the atmosphere can lead to increase of particle number/mass. NPF is one of the main sources of the atmospheric aerosols under the clean environment (such as forest (Makela et al., 1997) and Arctic marine boundary layer (Covert et al., 1996)). Binary nucleation of \( \text{H}_2\text{SO}_4-\text{H}_2\text{O} \) and ternary \( \text{H}_2\text{SO}_4-\text{H}_2\text{O}-\text{NH}_3 \) were considered as the commonly way for NPF (Seinfeld and Pandis, 2006). Recent study reported that the presence of amine (Kurten et al., 2008; Paasonen et al., 2012; Lv et al., 2015) or organic compounds (Zhang et al., 2004; Schobesberger et al., 2013; Wang et al., 2017) may play an important role in NPF events. NPF can lead to significant increase of ultrafine particle number concentration (Kulmala et al., 2004), however, further studies are needed to quantify its contribution to total particle mass loading. Secondary organic aerosols (SOA) are major contributors to aerosol mass loading, as reported by previous measurements (Zhang et al., 2007; Jimenez et al., 2009; Huang et al., 2014). The common mechanism of SOA formation is the oxidation of volatile organic compounds (VOCs), forming products of lower volatility that subsequently condense to the pre-existing particles. Organic aerosols account for a significant fraction of atmospheric aerosols (Saxena and Hildemann, 1996; Kanakidou...
et al., 2005; Zhang et al., 2007). There is growing evidence to show that the organic aerosols, especially SOA particles, play important role in affecting the air quality. On the other hand, secondary inorganic aerosols (SIA), mainly nitrate and sulfate, constitute also a major contributor to particle mass loading, as reported by previous observational (Liu et al., 2017; Wu et al., 2018) and modelling studies (Chen et al., 2016a; Chen et al., 2018). SIA is usually formed from the oxidation of SO\textsubscript{2} and NO\textsubscript{x} in heterogeneous reactions (Lelieveld and Heintzenberg, 1992; Riemer et al., 2003; Brown and Stutz, 2012; Chen et al., 2016a; Chen et al., 2018) and gaseous reactions followed by neutralization with ammonia and condense on pre-existing particle surface (Seinfeld and Pandis, 2006). Therefore, the oxidation of VOCs is a critical process of SOA formation, the oxidation of SO\textsubscript{2} and NO\textsubscript{x} are an important process of SIA formation.

Photochemical reactions are important source of oxidizing agents in the atmosphere, e.g., ozone and OH radical. Aerosol optical properties are suggested to be good indicators of photochemical reaction events. Aerosol optical properties are believed to be related to their chemical composition. It is well known that light absorption by dust and black carbon aerosols has a warming effect on climate (Jacobson, 2001; Yang et al., 2009; Bond et al., 2013) while light scattering from inorganic aerosols such as sulfate, nitrate, and sea salt has a cooling effect. Field and laboratory studies have reported that the optical properties of organic aerosols are very complex and most
of them exhibit light absorption properties. Chow et al. (2009) observed increasing light absorption toward shorter wavelengths during afternoon hours at the Fresno supersite when the SOA formed by photochemical reactions dominated the total aerosol carbon mass. All those studies suggest that the aerosol absorption related parameters, such as absorption coefficient ($b_{abs}$) and especially the aerosol concentration independent parameter single scattering albedo (SSA), are good indicators of the atmospheric photochemical reaction events.

In the last decade, aerosol optical properties have been investigated intensively in the world (Table 1), especially in China, but only limited studies were conducted in Europe, and generally those limited works were most concentrated in the Mediterranean area (see Table 1). In March 2014, a heavy pollution episode has occurred in the central interior of France, the observed PM$_{10}$ concentration in Paris was higher than its “alert level”, i.e. 80µg/m$^3$. The central interior of France is under the influence of both polluted air masses from industrial countries (Eastern Europe) and clean ones from Atlantic Ocean (Birmili et al., 2001; Hu et al., 2014). The mixing of these different types of air masses may lead to complex aerosol optical properties in this region. Our previous studies (Hu et al., 2014) have measured the aerosol extinction coefficient in this region (Orleans, France) during autumn time in 2012. The average aerosol extinction coefficient ($b_{ext}$) at 532 nm measured was 36 ± 31 Mm$^{-1}$, about 4 times higher than the
values of the air masses from Atlantic Ocean. However, up to now, the knowledge of aerosol optical properties, especially SSA, in the central region of France is very limited. Moreover, the contribution of air masses from Atlantic Ocean and industrial countries (Eastern Europe) to the atmospheric aerosols in the central interior of France is still unclear.

Table 1 Summary of the reported atmospheric aerosol optical property measurement studies.

In this study, aerosol scattering and extinction coefficient were measured in Orleans, a central region of France, during spring and autumn time of 2013. Simultaneous measurements of PM$_{10}$ and gaseous pollutants such as NO$_x$ and O$_3$ were also conducted during this period. The influences of air masses from Atlantic Ocean and industrial countries on the air quality of the central region of France were explored based on those data and the air masses back trajectory analysis. In addition, the contribution of photochemical reaction process to the atmosphere aerosol mass concentration was estimated according to the data of particle size distribution, boundary layer, human activity factor and PM$_{10}$ concentration.

2. Experimental setup and methods
2.1. Sampling site and period

The measurements presented in this study were made during the period from 7 March to 25 April and from 25 October to 5 December, 2013 at the campus of the Centre National de la Recherche Scientifique (CNRS) in Orleans, France (47°50′16.80″N, 1°56′39.34″E). Orleans city is located in the central France, about 120 km southwest of Paris. The typical average concentrations of O$_3$, NO$_2$, NO and PM$_{10}$ from May 2012 to March 2013 in Orleans are 52.8 ± 12.7, 24.0 ± 14.4, 4.3 ± 3.4 and 18.0 ± 5.9 μg/m$^3$, respectively (http://www.ligair.fr/). Our sampling site, positioned at the CNRS campus, is mostly surrounded by residential dwellings and about 8 km south of the Orleans city center. There are no obstructing buildings around the sampling site within 50 m. During the measurement, Cavity Ring-down Spectrometer (CRDS) and nephelometer (Model 3563, TSI, USA) sampled the ambient aerosols from an inlet located 1 m above the roof of the site (∼3.5 m above the ground). Before entering the instruments, the ambient aerosols passed through a 2 m stainless steel tube and a silica drier. In this study, the desiccant was replaced 2 times per week and the relative humidity (RH) of the sample before and after changing the desiccant was monitored for each replacement. The recorded data reveal that the RH of the dehydrated aerosol is less than 15%. Particle number size distribution was also measured by using a scanning mobility particle sizer (SMPS) from 31 October to 9 November, 2013.
2.2. Aerosol Optical Properties

To quantify aerosol optical effects, aerosol optical properties have commonly been determined by measuring scattering coefficient ($b_{\text{sca}}$) using integrated nephelometers (Heintzenberg and Charlson, 1996; Li et al., 2011; Esteve et al., 2012; Li et al., 2013), measuring absorption coefficients ($b_{\text{abs}}$) using the particle/soot absorption photometer (PSAP) (Vrekoussis et al., 2005; Li et al., 2007; Kalivitis et al., 2011), Aethalometer (Hansen et al., 1984; Arnott et al., 2005; He et al., 2009; Kalivitis et al., 2011) and photoacoustic spectrometer (PAS) (Andreae et al., 2008; Garland et al., 2009), as well as measuring extinction coefficient using cavity ring-down spectroscopy (CRDS) (Nakayama et al., 2010; Li et al., 2011; Li et al., 2013; Hu et al., 2014). According to those optical parameters, a useful variable to describe the relative effects of scattering compared with absorption, i.e., single scattering albedo (SSA), can be derived by the equation of $\text{SSA} = \frac{b_{\text{sca}}}{b_{\text{ext}}}$.

In this study, light extinction coefficients of ambient particles were measured using a CRDS at 532 nm. The CRDS system is a self-built and has been described in detail elsewhere (Hu et al., 2014). Briefly, laser light at 532 nm, generated by a pulsed Nd:YAG laser (EXPL-532-200-E, Spectra-Physics Inc., USA), is firstly introduced into a cavity cell and then bounced numerous times between two highly reflective mirrors (532 nm center wavelength, 99.997% reflectivity, 1m radius of curvature, ATFilms Inc.,
USA). Each time the pulse interacts with the back mirror, a small amount of light (e.g., 0.003%) leaks out and is detected by a photomultiplier tube (PMT, R928, Hamamatsu Inc., Japan). After getting the decay time of the cavity filled with ($\tau$) and without ($\tau_0$) aerosols, the aerosol extinction coefficient ($b_{ext}$) can be calculated according to the following equation:

$$b_{ext} = \frac{L}{cl} \left( \frac{1}{\tau} - \frac{1}{\tau_0} \right)$$  \hspace{1cm} (1)

where $L$ is the length between the mirrors (m), $l$ is the effective sample path length (m), $c$ is the speed of light (m/s). During the experiments, the CRDS was operated at the sample flow rate of 2 lpm (liter per minute), resulting a residence time of around 32 s. In addition, in order to avoid interferences from NO$_2$ and other absorbing trace gas species, blank checks of the CRDS system were performed automatically every 2 h by turning the 3-way valve to the filter direction allowing the measurement of $b_{ext}$ for the particle-free ambient air. Before each measurement, CRDS was calibrated by polystyrene latex spheres (PSL) particles.

Light scattering coefficients of ambient particles were measured using a three-wavelength (450, 550, 700 nm) integrating nephelometer (TSI model 3563, USA). In this study, it was operated at the sample flow rate of 2.5 lpm with a minute averaging time. To avoid the interference of the gas-phase compounds, a 10 min zero check was done automatically by measuring $b_{sca}$ of the particle-free ambient air every 2 h. Before
each field campaign, the nephelometer was calibrated in the laboratory using CO$_2$ and the filtered air. Due to the design limitations, the scattered light in the nephelometer was integrated over an angular range of 7-170°, and did not cover the full angular range (0-180°). Thus, a truncation correction developed by Anderson and Ogren (1998) was applied to the measurement data. In addition, to derive the SSA data, the scattering coefficient at 532 nm was calculated by the following equation:

$$b_{\text{sca},532} = b_{\text{sca},\lambda} \times \left( \frac{532}{\lambda} \right)^{-\tilde{\alpha}}$$  \hspace{1cm} (2)

where the Ångstrom exponent ($\tilde{\alpha}$) can be derived from the measurement of $b_{\text{sca},\lambda}$ at two different wavelengths $\lambda_1$ and $\lambda_2$ through the following equation:

$$\tilde{\alpha} = -\frac{\log(b_{\text{sca},\lambda_1} / b_{\text{sca},\lambda_2})}{\log(\lambda_1 / \lambda_2)}$$  \hspace{1cm} (3)

2.3 Particle Number Size Distribution

The SMPS was described in detail elsewhere (Heim et al., 2004). Herein, only a brief description is presented. In the SMPS system used in this work (TSI Inc., USA), the particles are firstly neutralized with Kr$^{85}$ diffusion charger and size selected using a differential mobility analyzer (DMA, Model 3081). The selected monodisperse particles subsequently go to a butanol-based condensation particle counter (CPC, Model 3022A), which determines the particle concentration at that size. In this study, the detectable aerosol mobility diameters ranged from 10.7 to 487.3 nm with a sheath...
and sample flow rates of 5 lpm and 0.5 lpm, respectively. The total counting time required for each sample was about 130 s, consisting of a 120 s up scan and a 10 s retrace. Before measurements, SMPS was calibrated with PSL particles.

2.4 Assessment of secondary particles

In this study, we calculated PM$_{10}$ concentration (PM$_{10\_calc}$) with respect of boundary layer dilution and fresh anthropogenic emissions from human activities (as described in the following) by using equation (4), and compare PM$_{10\_calc}$ with the measured PM$_{10}$ concentration (PM$_{10\_meas}$) to investigate the potential influence of secondary particles formation on PM loading.

$$PM_{10\_calc} = PM_{10\_meas} \times f(\text{anthropogenic emission}) \times f(\text{BLD})$$ (4)

where $f(\text{anthropogenic emission})$ is the human activity factor in term of PM emission which was derived from the anthropogenic emission flux of PM at the measurement site, and $f(\text{BLD})$ is the dilution factor of boundary layer change. Both factors are described below in detail.

The anthropogenic emission flux of PM at the measurement site (Kuenen et al., 2014) is presented in Fig S-2. The emission data from TNO (Nederlandse Organisatie voor Toegepast Natuurwetenschappelijk Onderzoek) with consideration of emissions from industry, traffic, residential, agriculture, power and energy and etc. In this study,
we assume the particle loading during the period from 00:00 to 07:00 (local time) is influenced by the anthropogenic emissions (Fig S-1) and photochemical reaction very slightly (PM$_{10}$ concentration was stable during this time period which is showed by the red curves in Fig.2.). While after 07:00, these emissions become stronger and could be a main source of ambient aerosols. To better understand the contribution of anthropogenic emission to the ambient aerosol during the day time, anthropogenic emission flux during 00:00-07:00 were chosen as the reference and normalized to 1, the human activity factor ($f$(anthropogenic emission)) in term of PM emission during the rest hours (07:00-00:00) is calculated as a ratio to the reference value, assuming that the fresh emitted PM is mixed in the boundary layer during the rest hours of the day.

The dilution effect of boundary layer mixing is also considered as following. During the night (00:00-07:00), PM$_{10}$ and Boundary layer depth (BLD) are stable for both campaigns (Fig. 2). The data in this time period are averaged and used as the reference to calculate the variation of PM$_{10}$ concentration which is only due to the dilution effect of boundary layer changing for the remaining hours (07:00-00:00).

3. Results and discussion

3.1 Impact of air mass origination on air quality

Fig. 1 illustrates the $b_{ext}$, $b_{scat}$, $b_{abs}$, SSA at 532 nm, PM$_{10}$, O$_3$, NO, NO$_2$, NO$_x$, RH and temperature (T) throughout the measurement campaign during the spring time from
7 March to 25 April and during the autumn time from 25 October to 5 December 2013, the corresponding averaged values are summarized in Table 2. The different terms appearing in the second and third rows of Table 2 are further defined and explained in the text. During the whole measurements, the particle optical parameters such as $b_{\text{ext}}$, $b_{\text{sca}}$ and $b_{\text{abs}}$ are in good agreement with the PM$_{10}$ concentration ($R^2 = 0.95$, 0.95 and 0.75, respectively). For the spring campaign (Fig. 1(a)), the average values of $b_{\text{ext}}$, $b_{\text{sca}}$ and $b_{\text{abs}}$ at 532 nm were $65.6 \pm 67.8$ Mm$^{-1}$, $54.9 \pm 58.2$ Mm$^{-1}$ and $10.6 \pm 10.9$ Mm$^{-1}$, respectively. This result is comparable with those observed in Eastern and Western Mediterranean (as shown in Table 1), but much lower than those observed in the Mega cities of China (Beijing, Shanghai and Guangzhou) and Japan (Tokyo). During this period, two pollution episodes were observed from 24 March to 1 April (represented as “pollution days”, noted as “P-1”) and from 2 April to 7 April (noted as “P-2”). The maximum values of PM$_{10}$ and $b_{\text{ext}}$ during “P-1” and “P-2” periods reached up to 93.4 μg/m$^3$, 317.8 Mm$^{-1}$ and 73.8 μg/m$^3$, 249.1 Mm$^{-1}$, respectively. It is noteworthy that (see Fig. 1(a)) the northeast (NE) wind was dominant during these two pollution events and the air parcel was mainly coming from Eastern Europe (Fig. S-1, calculated by HYSPLIT trajectory model, NOAA, http://ready.arl.noaa.gov/HYSPLIT.php). This result suggests that the air parcel transported from Eastern Europe may be responsible for those two pollution events, a similar event is reported over Germany (Chen et al.,
2016b). As shown in Fig. 1(a), the air quality during the period from 16 March to 19 March (represented as “clean days”, noted as “C-1”) and from 9 April to 13 April (noted as “C-2”) is much better than during the other days. The average PM$_{10}$ and $b_{ext}$ for “C-1” and “C-2” periods are only 10.0 $\mu g/m^3$, 23.5 Mm$^{-1}$ and 11.7 $\mu g/m^3$, 10.1 Mm$^{-1}$, respectively. Meteorology (Fig. 1(a)) and air parcel back trajectory (Fig. S-1(a)) reveal that the wind direction during the period of “C-1” and “C-2” was almost southwest (SW), and the air parcel was mainly coming from the Atlantic Ocean. This result is consistent with our previous observations (Hu et al., 2014), i.e. air parcel transported from Atlantic Ocean could improve the air quality of Orleans. In addition, to better understand the influence of air masses from Atlantic Ocean and Eastern Europe on the air quality in Orleans, the period from 20 March to 23 March (noted as “L-1”) was selected to represent the local pollution conditions in Orleans since the wind speed is almost zero during this period. As can be seen in Table 2, the average PM$_{10}$ and $b_{ext}$ values during “L-1” period are 30.4 $\mu g/m^3$ and 77.9 Mm$^{-1}$, respectively, about 2.0 and 2.1 times lower than “P-1”, 1.5 and 1.7 times lower than “P-2”, while about 3.0 and 3.3 times higher than “C-1”, 2.6 and 7.7 times higher than “C-2”.

**Fig. 1. Time series of $b_{ext}$, $b_{sca}$, $b_{abs}$, SSA at 532 nm, PM$_{10}$, O$_3$, NO, NO$_2$, NO$_x$, RH and T from: (a) 7 March to 25 April, (b) 25 October to 5 December 2013.**
Table 2. Summary of the average and maximum values of $b_{\text{ext}}$, $b_{\text{sca}}$, $b_{\text{abs}}$, SSA, PM$_{10}$, O$_3$, NO, NO$_2$ and NO$_x$ during the different period.

For the autumn campaign (Fig. 1(b)), the average values of $b_{\text{ext}}$, $b_{\text{sca}}$, $b_{\text{abs}}$ at 532 nm and PM$_{10}$ were $37.5 \pm 37.8$ Mm$^{-1}$, $35.4 \pm 36.7$ Mm$^{-1}$, $3.9 \pm 4.4$ Mm$^{-1}$ and $17.4 \pm 11.8$ μg/m$^3$ respectively, about 2 times lower than the values observed during the spring campaign. As shown in Fig. 1(b), the air quality during the period from 25 October to 9 November, represented as “clean days” and noted as “C”, was much better than other days. The averages of PM$_{10}$ and $b_{\text{ext}}$ during the “C” period were only $3.8 \, \mu g/m^3$ and 7.5 Mm$^{-1}$, respectively. As expected, the SW wind was dominant (Fig. 1(b)) for those clean days and the air parcel was coming from Atlantic Ocean (Fig. S-1(b)). While for other days, from 10 November to 5 December (noted as “M”), the wind direction (Fig. 1(b)) was complex and the air masses came from several different sources (Fig. S-1(b)). The mixing of these different types of air masses can be representative of the typical atmosphere mixing conditions in the Orleans area. As shown in Table 2, the average PM$_{10}$ and $b_{\text{ext}}$ values during the “M” period were $22.0 \pm 12.7$ μg/m$^3$ and $52.7 \pm 40.5$ Mm$^{-1}$, respectively, about 1.3 and 1.4 times larger than the average values during the whole autumn campaign and 2.2 and 4.3 times larger than during “C” period.
3.2 Diurnal variations of PM$_{10}$, O$_3$, NO$_x$ and particles optical properties

Fig. 2 shows the statistic diurnal variation of the PM$_{10}$, BLD, NO$_x$, O$_3$ and photolysis rate of NO$_2$ ($J_{NO2}$) during the spring and autumn time. In this “box” diagrams, the mean value is represented by a dot. The dividing segment in the box is the median. The top and bottom box limits represent the 75 and 25 percentile, respectively. The error bars are related to the percentiles 95th and 5th. As shown in Fig 2, the BLD (calculated by Archived Meteorology, GDAS (1 deg. 3hourly. Global), NOAA, http://ready.arl.noaa.gov/READYamet.php) is stable at night while increases consistently in the daytime. The maxima of the average BLD in the daytime are around 1200 and 900 m for the spring and autumn campaigns, respectively, about 5 and 2 times larger than the corresponding night time values. Since the particles concentration is highly influenced by the BLD, the fast increases of BLD in the daytime should decrease the PM$_{10}$ concentration quickly. However, in the real conditions of our measurements (Fig. 2), the daily PM$_{10}$ concentration is stable and only shows a slight decrease during afternoon hours. A reasonable explanation for this phenomenon is that the formed or produced particles from the atmospheric chemical reactions in the daytime may counteract the dilution effect due to the variation of the BLD.
Fig. 2. Statistic diurnal variation of PM$_{10}$, Boundary layer depth (BLD), NO$_x$, O$_3$ and J$_{NO2}$ from: (a) 7 March to 25 April, (b) 25 October to 5 December 2013.

Several field studies have shown that atmospheric photochemical reactions are the main source of atmospheric particles in the daytime. In this study (Fig. 2), a distinct diurnal photochemical reaction process, i.e. strong O$_3$ formation, was observed in both campaigns. Moreover, there are limited industries in the central region of France and our sampling site is mostly surrounded by residential dwellings, the photochemical reaction processes should be proposed to be the main source of ambient aerosols in Orleans. To further support this point, the statistic diurnal variation of aerosols optical properties (including $b_{abs}$, $b_{sca}$ and SSA) during the spring and autumn time are present in Fig. 3. In these figures, the SSA exhibits a clear daily pattern, with three “drop down” within a day at 8:00, 14:00 and 18:00 for the spring campaign and at 8:00, 12:00 and 17:00 for the autumn campaign. The daily variability of SSA occurs in the morning, and in the evening is mostly due to the rush hours (Esteve et al., 2012). There is a large amount of black carbon particles emitted during this period. Whereas for the SSA variation occurs in the early afternoon, it is supposed to be derived by the SOA which are formed through photochemical reaction processes because: (1) Field and laboratory studies (Andreae and Gelencser, 2006; Moise et al., 2015) reveal that the organic
aerosols formed by photochemical reactions are another light-absorber in the atmosphere apart from black carbon and mineral dust; (2) the SSA second “drop down” appears at the highest solar intensity (highest JNO2) time. The peak of JNO2 is measured at 13:00 and 12:00 within a day for the spring and autumn campaigns, respectively.

Fig. 3. Statistic diurnal variation of b_{abs}, b_{sca} and SSA at 532 nm from: (a) 7 March to 25 April, (b) 25 October to 5 December 2013.

To better understand the impact of photochemical reactions on the air quality in the central region of France, the contribution of photochemical reaction processes to the atmosphere aerosols needs to be estimated. As shown in Fig.2, the PM10 and BLD are stable during the night (00:00-07:00) for both campaigns. We assume the variation of PM10 concentration is only due to the dilution effect (BLD changes) for the remaining hours (07:00-00:00) and can be calculated according to this stable condition. The difference of PM10 between the calculated and measured values can be considered as the contribution of the atmosphere photochemical reaction processes. Actually, this value was overestimated because not only the photochemical reaction processes, but other processes like emissions from human activities can also contribute to the atmospheric aerosol pollution. To avoid the influence of this, the human activity factor
derived from anthropogenic emission rate of PM$_{2.5}$ (from TNO-MACC-II inventory, (Kuenen et al., 2014)) was considered in this work (as shown in Fig. S2). The calculated variation of PM$_{10}$ concentration which considered both BLD dilution and human activities effects is presented as the green curve in Figure 2 (the detailed calculation and parameters was described in part 2.4). Based on the above estimations, the fraction of the photochemical reaction formed particles in PM$_{10}$ was calculated ((PM$_{10, \text{meas}}$ – PM$_{10, \text{calc}}$) / PM$_{10, \text{meas}}$) and presented in Fig.4. It reveals that particles formed from photochemical reactions can account maximum of 64% and 32% for PM$_{10}$ concentration during the spring and autumn times, respectively. These results indicate that photochemical reactions may play an important role in impacting air quality in Orleans.

It is worth to clarify that, in this section, we use the statistical analysis, which based on the data during the whole period in spring or autumn, to show the photochemical reactions can enhance the atmospheric oxidation capacity and may faster the secondary particle formation and then play an important role in air quality. But in section 3.1, we give the case studies to show the air parcel transported from Atlantic Ocean plays a role in cleaning up the ambient air in Orleans, while the air mass coming from the Eastern Europe induces the pollution events in Orleans. Those episodes are occurred with
limited times and short period. The results get from those 2 different analysis methods are not contradictory.

Fig. 4. The estimated fraction of photochemical reaction producing particles in $\text{PM}_{10}$.

3.3. Particles formation and growth events during the typical photochemical reaction process.

To better understand the influence of atmospheric photochemical reaction processes on the air quality in Orleans, the statistic diurnal variation of particles size distribution, NO$_x$, O$_3$ and J$_{NO2}$ during the period from 31 October to 9 November, 2013 are presented in Fig. 5. As shown in Fig. 1(b), the air quality is much better during these days. The southwestern wind is dominant during this period and the air parcel comes from Atlantic Ocean (Fig. S-1(b)). This atmospheric meteorological condition is suitable for the observation of new particle formation and growth events since particles transported from Atlantic Ocean are mainly of the micrometer-level (sea salt aerosols). This latter can be easily distinguished with the newly formed nanometer-level particles.
Fig. 5. Statistic diurnal variation of the particle size distribution, NO$_x$, O$_3$ and J$_{NO2}$ from 31 October to 9 November, 2013.

As shown in Fig. 5, a clear new particles formation process was observed around 13:00, the maximum particle number concentration reached up to 3488 #/cm$^3$. As expected, a typical photochemical reaction process with O$_3$ formation was observed during this period, and O$_3$ approached its peak value of 60 µg/m$^3$ (median and mean values) around 13:00. This suggests that photochemical reactions enhance the new particle formation, which may play an important role in impacting air quality in Orleans. Furthermore, the highly humid condition (RH = 92 % in average) during this period also favors the new particle formation process.”

4. Conclusions

In this study, the aerosols optical properties, size distribution and PM$_{10}$ concentration measurements were conducted at Orleans, France during spring (7 March to 25 April) and autumn (25 October to 5 December) 2013. The average values of $b_{sca}$, $b_{abs}$, SSA at 532 nm and PM$_{10}$ concentration are $54.9 \pm 58.2$ Mm$^{-1}$, $10.6 \pm 10.9$ Mm$^{-1}$, $0.81 \pm 0.10$, $30.6 \pm 21.6$ µg/m$^3$ and $35.4 \pm 36.7$ Mm$^{-1}$, $3.9 \pm 4.4$ Mm$^{-1}$, $0.83 \pm 0.13$, $17.4 \pm 11.8$ µg/m$^3$ for the spring and autumn, respectively. During the whole observation, the air parcel transported from Atlantic Ocean plays a role in cleaning up the ambient air
in Orleans, while the air mass coming from the Eastern Europe induces pollution events in Orleans. In addition, a distinct diurnal pattern of photochemical reaction and ozone formation were observed in pollution cases. Further analysis reveals that particles formation triggered by photochemical reactions in daytime can account for a maximum of 64% and 32% of PM$_{10}$ mass loading during the spring and autumn, respectively. Our results highlight that photochemical reactions may play an important role in secondary particles formation and new particles formation and then impacting air quality in the central region of France, and possibly other regions as well.

Acknowledgment

This work was supported by FP7 project (AMIS, No.PIRSES-GA-2011), ARD PIVOTS program (supported by the Centre-Val de Loire regional council), Labex VOLTAIRE (ANR-10-LABX-100-01), the National Natural Science Foundation of China (Nos. 41305124). Dawei Hu thanks Shanghai Tongji Gao Tingyao Environmental Science & Technology Development Foundation for a fellowship support.

References


Schobesberger, S., Junninen, H., Bianchi, F., Lonn, G., Ehn, M., Lehtipalo, K., et al., 2013. Molecular understanding of atmospheric particle formation from sulfuric acid


atmospheric measurements during EAST-AIRE. Atmospheric Chemistry and Physics. 9, 2035-2050.


Table 1 Summary of the reported atmospheric aerosol optical property measurement studies.

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<th>Country</th>
<th>Site</th>
<th>Period</th>
<th>$b_{sc}$ (Mm$^{-1}$)</th>
<th>$b_{abs}$ (Mm$^{-1}$)</th>
<th>SSA</th>
<th>Instruments</th>
<th>Reference</th>
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</thead>
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<td>China</td>
<td>Shanghai</td>
<td>2009.12.23-2009.12.25</td>
<td>591 ± 143 (532 nm)</td>
<td>201 ± 51 (532 nm)</td>
<td>~ 0.75</td>
<td>CRDS Nephelometer</td>
<td>Tang et al. (2014)</td>
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<td>2009.11.27-2009.12.06</td>
<td>332 ± 217 (532 nm)</td>
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<td>Li et al. (2011)</td>
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<td>Shanghai</td>
<td>2010.04.02-2010.05.05</td>
<td>102 ± 75 (532 nm)</td>
<td>44 ± 35 (532 nm)</td>
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<td>CRDS Nephelometer</td>
<td>Li et al. (2013)</td>
</tr>
<tr>
<td></td>
<td>Beijing</td>
<td>2012.05.24-2012.06.22</td>
<td>/</td>
<td>/</td>
<td>~ 0.80 (550 nm)</td>
<td>CRDS Nephelometer Aethalometer</td>
<td>Li et al. (2017)</td>
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<tr>
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<td>2005.01-2006.12</td>
<td>288 ± 281 (525 nm)</td>
<td>56 ± 49 (532 nm)</td>
<td>0.80 ± 0.09</td>
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<td>He et al. (2009)</td>
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<td>2006.08.11-2006.09.09</td>
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<td>2005.03</td>
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Table 2 Summary of the average and maximum values of $b_{\text{ext}}$, $b_{\text{sca}}$, $b_{\text{abs}}$, SSA, PM$_{10}$, O$_3$, NO, NO$_2$ and NO$_x$ during the different period.

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Figure Captions

**Fig. 1.** Time series of $b_{\text{ext}}$, $b_{\text{sca}}$, $b_{\text{abs}}$, SSA at 532 nm, PM$_{10}$, O$_3$, NO, NO$_2$, NO$_x$, RH and T from: (a) 7 March to 25 April, (b) 25 October to 5 December 2013.

**Fig. 2.** Statistic diurnal variation of PM$_{10}$, Boundary layer depth (BLD), NO$_x$, O$_3$ and $J_{\text{NO2}}$ from: (a) 7 March to 25 April, (b) 25 October to 5 December 2013.

**Fig. 3.** Statistic diurnal variation of $b_{\text{abs}}$, $b_{\text{sca}}$ and SSA at 532 nm from: (a) 7 March to 25 April, (b) 25 October to 5 December 2013.

**Fig. 4.** The estimated fraction of photochemical reaction producing particles in PM$_{10}$.

**Fig. 5.** Statistic diurnal variation of the particle size distribution, NO$_x$, O$_3$ and $J_{\text{NO2}}$ from 31 October to 9 November, 2013.
Fig. 1
Fig. 2
Fig. 3

(a) Absorption (Mm$^{-1}$) vs. Local Time (h)

(b) Scattering (Mm$^{-1}$) vs. Local Time (h)

(c) SSA (single scattering albedo) vs. Local Time (h)
Fig. 5