



HAL
open science

A new Eemian record of Antarctic tephra layers retrieved from the Talos Dome ice core (Northern Victoria Land)

Biancamaria Narcisi, Jean Robert Petit, Antonio Langone, Barbara Stenni

► To cite this version:

Biancamaria Narcisi, Jean Robert Petit, Antonio Langone, Barbara Stenni. A new Eemian record of Antarctic tephra layers retrieved from the Talos Dome ice core (Northern Victoria Land). *Global and Planetary Change*, 2016, 137, pp.69 - 78. 10.1016/j.gloplacha.2015.12.016 . insu-01351730

HAL Id: insu-01351730

<https://insu.hal.science/insu-01351730>

Submitted on 9 Mar 2021

HAL is a multi-disciplinary open access archive for the deposit and dissemination of scientific research documents, whether they are published or not. The documents may come from teaching and research institutions in France or abroad, or from public or private research centers.

L'archive ouverte pluridisciplinaire **HAL**, est destinée au dépôt et à la diffusion de documents scientifiques de niveau recherche, publiés ou non, émanant des établissements d'enseignement et de recherche français ou étrangers, des laboratoires publics ou privés.

1 **A new reconstruction of atmospheric gaseous elemental**
2 **mercury trend over the last 60 years from Greenland firn**
3 **records**

4 **A. Dommergue¹, P. Martinerie¹, J. Courteaud¹, E. Witrant², D. M. Etheridge³**

5 [1] Univ. Grenoble Alpes/CNRS, Laboratoire de Glaciologie et Géophysique de
6 l'Environnement (LGGE) UMR 5183, Grenoble, F-38041, France

7 [2] Univ. Grenoble Alpes/CNRS, Grenoble Image Parole Signal Automatique (GIPSA-lab),
8 UMR 5216, B.P. 46, F-38402 St Martin d'Hères, France

9 [3] CSIRO Oceans and Atmosphere, 107-121 Station St., Aspendale, Victoria 3195,
10 Australia

11 Correspondence to: Aurélien Dommergue (aurelien.dommergue@univ-grenoble-alpes.fr)

12 **Abstract**

13 This study presents measurements of gaseous elemental mercury (GEM) concentrations in the
14 80 meters of firn air at the international drilling site of NEEM in Greenland (2452 m, 77°25.8
15 N, 51°06.4 W). Using inverse modeling, we were able to reconstruct the atmospheric GEM
16 trend at this Arctic site over the last 60 years. We show discrepancies between this record and
17 the previous firn record of Summit. This could be attributed to experimental biases and/or
18 differences in air mass transport. A multisite inverse model was used to derive an atmospheric
19 scenario reconciling the two firn records. We show that GEM seasonal variations are very
20 limited at these high altitude sites and thus probably unaffected by spring/summer
21 photochemistry. The firn reconstructions suggest an increase of GEM concentrations since the
22 1950s peaking in the late 1960s and early 1970s. A decrease is then observed with minimum
23 GEM concentrations around 1995-2000. The reconstruction compares well with historical
24 mercury (Hg) releases and recent simulations of atmospheric Hg. Our optimal GEM scenario
25 does not allow to categorically conclude on recent trends for GEM concentrations over the
26 2000-2010 decade.

27

28

1 **1 Introduction**

2 The Arctic troposphere receives mercury (Hg) through air mass transporting anthropogenic
3 emissions (mining, fuel combustions) from Asian, Northern American and European sources
4 (Durnford et al., 2010). Hg mainly reaches the arctic troposphere as gaseous elemental
5 mercury (GEM), as it has a longer atmospheric lifetime than other Hg species. GEM is also
6 emitted by evasion from the Arctic Ocean (Dastoor and Durnford, 2013) and by re-emission
7 of deposited Hg from snow and ice surfaces (Durnford and Dastoor, 2011). The oceanic Hg
8 source, fed by circumpolar riverine inputs, was recently shown to dominate at coastal Arctic
9 sites, where it could cause a summertime atmospheric GEM maximum (Fisher et al., 2012).

10 Atmospheric GEM concentrations in the Arctic troposphere are one of the key parameters that
11 controls Hg deposition. How the Arctic atmosphere responded to historical changes in
12 anthropogenic emissions is an important question to answer in order to understand the
13 present-day biogeochemical cycle of Hg in the Arctic. This understanding is also a
14 prerequisite for predicting how the Hg cycle will be affected by future Hg emission policies
15 (e.g. the Minamata Convention).

16 The longest monitoring record of atmospheric GEM in the Arctic began in 1995 at Alert
17 (82°N, Canada) and revealed important seasonal variations (Cole and Steffen, 2010; Steffen et
18 al., 2008). The High Arctic springtime atmosphere is characterized by extremely low GEM
19 values ($<1 \text{ ng.m}^{-3}$) due to a photochemically-driven rapid oxidation process involving
20 bromine, while summer months exhibit a return to higher GEM values ($\sim 2 \text{ ng.m}^{-3}$). The so-
21 called springtime atmospheric mercury depletion events (AMDE) are an important pathway
22 by which atmospheric GEM can be converted into more labile species (Larose et al., 2011). It
23 delivers large quantities of oxidized mercury (Hg(II)) on environmental surfaces for a short
24 period of time (Douglas et al., 2012). Hg(II) species are susceptible to photoreduction and a
25 fraction – which is large in the case of seasonal terrestrial snowpacks – is rapidly reemitted
26 back to the atmosphere as GEM (Poulain et al., 2004). The air monitoring records at Alert and
27 at Zeppelin station (79°N, Svalbard) show either slowly declining or stable GEM levels in the
28 past decade (Berg et al., 2008; Berg et al., 2013; Cole and Steffen, 2010; Cole et al., 2013).
29 These observations, as well as others from sub-Arctic sites (Cole et al., 2013), appear to
30 contradict rising global Hg emission trends (Streets et al., 2011). However, recent modeling
31 studies have highlighted the potentially important role played by the ocean (Chen et al., 2015;
32 Soerensen et al., 2012) in modulating atmospheric GEM levels (Fisher et al., 2013).

1 An indirect method to reconstruct long-term trends in atmospheric gases consists of
2 measuring their concentration profile in the interstitial air (also called firn air) of the upper
3 layers of polar ice sheets (e.g. Schwander et al., 1993; Witrant et al., 2012). Applying this
4 method at Summit in central Greenland, Faïn et al. (2009) reconstructed a history of Arctic
5 GEM levels over 60 years. Their measurements revealed that atmospheric GEM
6 concentrations peaked in the 1970s and were in line with global Hg production. In this paper,
7 we present GEM measurements obtained in the upper 80 m of the northeastern Greenland ice
8 sheet cap at the international drilling site NEEM (North Greenland Eemian Ice Drilling).
9 Using a new and robust modeling approach, we are able to propose a new atmospheric
10 scenario of GEM for the last 60 years. GEM data obtained at Summit (Faïn et al., 2009) are
11 re-evaluated and compared to our results.

12 **2 Firn air sampling and GEM analysis**

13 GEM measurements were carried out during the NEEM program from July 16th to July 29th,
14 2009. The NEEM site is situated in northeastern Greenland (Figure 1) at an altitude of 2452
15 meters above sea level. Experiments were conducted 1.8 km away from the main camp
16 (77°25.8 N, 51°06.4 W).

17 The sampling of firn air was first documented by Schwander et al. (1993). Here we describe a
18 modified setup for the extraction of large samples using a Firn Air Sampling Device (FASD).
19 Drilling progressed stepwise with sampling at intervals of 10 m until the beginning of the
20 lock-in zone (at ~60 m, (Buizert et al., 2012)). The lock-in zone is the firn region where air
21 bubbles begin to be trapped in the ice matrix and vertical gas diffusion significantly slows and
22 eventually stops. In the lock-in zone, drilling was made at intervals of 1 to 3 m. At each
23 sampling level the drill was withdrawn and the hole sealed close to the bottom with a 3 m
24 long inflatable butyl rubber bladder. A pumping system from CSIRO (KNF N286.15
25 Diaphragm compressor, 15-25 l/min) inflated the bladder to 90-150 kPa with air drawn from
26 the firn and monitored its pressure. Two continuous, 100 m long, 3/8 inch Nylon (type 12;
27 Watsford Tubetech, Nunawading, Victoria, Australia) tubes passed through the bladder,
28 connecting the pumping system at the surface to the bottom of the hole. Sampled air was
29 drawn through both tubes and flasks were filled for later trace gas concentrations and isotopic
30 ratios analysis. The quality of the sampling was checked at each level after inflating the
31 bladder, and at the end of the measurements. A CO₂ analyzer (Sick-Maihak 710) was used to
32 detect possible leaks from the bladder or tubes, and to verify the complete removal of modern

1 air introduced in the hole when drilling and lowering the bladder. It also provided an
2 indication of the mean age of the air, and whether air flowed from adjacent firn layers (with
3 different ages) during pumping. The variations in CO₂ concentrations at each level between
4 measurements were relatively small, typically decreasing by less than 2 ppm_v during sampling
5 at each level, despite pumping a total air volume of 1000-2000 L.

6 GEM concentrations in the firn air were measured online. An additional Teflon pump
7 (Vacuumbrand MZ2NT) and the GEM analyzer were connected to the FASD. System
8 components were tested in the lab during construction of the FASD for their effect on GEM,
9 for instance the Nylon tubing, which was found to have no effect. All components were steam
10 cleaned and baked in an oven before assembly. These parts of the system were leak-tested,
11 and several blanks of the FASD line were realized using GEM-free air produced by an
12 activated charcoal cartridge. Two sources of contamination were noticed during blank tests at
13 NEEM before proceeding: a needle valve, which was subsequently removed, and the Nylon
14 tubing when exposed to direct sunlight. Shading the tubing reduced the GEM blanks to below
15 detectable levels. The system used during this campaign did not show any evidence of GEM
16 contamination, while the previous study of Summit (Faïn et al., 2009) had to account for
17 positive blanks, that were subtracted from the measurements. GEM collection and analysis
18 were performed using a 2537A gas phase Hg analyzer (Tekran® Instruments Corporation).
19 This instrument uses gold-amalgamation, thermal desorption in a pure argon stream and Hg
20 detection by Atomic Fluorescence Spectrometry (AFS) at 253.7 nm. Two parallel gold
21 cartridges allow alternate sampling and desorption, resulting in continuous measurement of
22 GEM on a pre-defined time base. The analyzer was operated with a 5-min sampling frequency
23 and the pre-filtered (0.2 µm PTFE filters) air was sampled at a flow rate of 1 L.min⁻¹. At each
24 sampling depth, we continuously measured GEM for at least 55 min and up to 150 min
25 depending on the field schedule. GEM concentrations in the firn were very reproducible at a
26 given depth, with a relative standard deviation less than 5% (except at depth 73 m and 75 m
27 where the standard deviations were 9% and 7% respectively). The analyzer was calibrated
28 before each measurement using the internal permeation source. An external calibration with
29 Hg vapor injections was performed in the laboratory before the field work. Under these
30 conditions, the relative uncertainty of the instrument on one measurement was better than 0.1
31 ng.m⁻³.

1 GEM concentration at two meters above the snow surface were periodically measured during
2 the firn air experiment with a 5-min resolution. After the firn air experiment, continuous
3 atmospheric GEM measurements were made for 6 days providing an estimate of mean surface
4 levels.

5 **3 Firn air modeling for mercury**

6 **3.1 Laboratoire de Glaciologie et Géophysique de l'Environnement and** 7 **Grenoble Image Parole Signal Automatique (LGGE-GIPSA) model**

8 Firn is an open porosity medium where atmospheric gases move mainly by diffusion. Its
9 density increases from the surface to the firn-ice transition, and the firn diffusivity decreases
10 in consequence. Variations of atmospheric GEM concentrations occurring at the surface
11 propagate into the firn by molecular diffusion, gravitational settling and advection (due to the
12 fact that the firn sinks progressively as snow accumulates). The physics of our model are
13 described in detail in Rommelaere et al. (1997) and Witrant et al. (2012). Model evaluation
14 through an international model comparison is presented in Buizert et al. (2012). Site specific
15 description of firn diffusivity calculations can be found in Witrant et al. (2012) and in
16 Zuiderweg et al. (2013) for Summit and NEEM 2009 respectively. Using in turn the
17 diffusivity profiles tuned to trace gas and gas isotopic measurements at each of the firn air
18 pumping operations performed at NEEM in 2008 and 2009 (Note: GEM measurements are
19 only available for the NEEM 2009 experiment) had no significant influence on the results (see
20 supplementary Figure S4).

21 Two GEM specific physical constants are used in the gas transport model: its atomic
22 weight ($200.59 \text{ g}\cdot\text{mol}^{-1}$) and its relative diffusion coefficient in air with respect to CO_2 . Based
23 on GEM diffusion coefficient measurements (Lugg, 1968; Massman, 1999), CO_2 diffusion
24 coefficient measurements (Massman, 1998; Matsunaga et al., 1998) and the calculations from
25 Chen and Othmer (1962) discussed in Buizert et al. (2012), a range of 0.822-0.868 was
26 obtained for the GEM/ CO_2 diffusion coefficient ratio. The value of 0.868 derived from Chen
27 and Othmer (1962) is used below, and a sensitivity test using 0.822 had only a small impact
28 on the results (see supplementary Figure S4).

29 Inverse modeling of gas transport in the firn is required to reconstruct an atmospheric
30 time trend from concentration vertical profiles. It is a mathematically under-determined
31 problem, which means that it has several possible solutions (Rommelaere et al., 1997). We

1 use the most recent version of the LGGE-GIPSA model (Witrant and Martinerie, 2013) in
2 which the optimal solution is determined using the cross-validation statistical method. The
3 concentrations are forced to be positive using a log barrier method (Boyd and Vandenberghe,
4 2004). Imposed positivity of concentrations mostly improves the results for recently emitted
5 anthropogenic compounds with null concentrations in deep firn (e.g. Laube et al., 2014).
6 Recent reconstructions of CO (Petrenko et al., 2013) and light alkanes (Helmig et al., 2014)
7 with the LGGE-GIPSA model are interesting to compare with the present GEM
8 reconstructions because all these datasets show peak shapes in the deep firn, and a good
9 consistency between several Greenland sites was obtained for CO and light alkanes.

10 **3.2 Age distribution in the firn**

11 Due to mixing by molecular diffusion, the GEM atoms found at a given depth in the firn
12 have different ages. Using a Dirac function as input, the forward model calculates an age
13 distribution (or Green's function) for a given gas at each depth in the firn (Rommelaere et al.,
14 1997). Table 1 and Figure S1 (Supplementary Information) show the estimated mean age of
15 GEM ($\pm 1\sigma$) in the firn at NEEM and Summit. Ages increase slowly in the upper part of the
16 firn, then rapidly in the lock-in zone (where most of air bubbles close) and diffusion
17 effectively ceases. Diffusional smoothing results in increasing gas age distribution with depth,
18 mainly above the lock-in zone. The age distribution stabilizes in the lock-in zone where
19 diffusion becomes negligible and advection with the ice dominates. The overlap between
20 possible gas age distribution at two successive sampling depths (especially in deep firn)
21 implies that GEM measurements are auto-correlated and that the firn acts as low-pass filter.
22 The model interprets irregular concentration variations in firn either as large variations in the
23 atmosphere or as noise. The optimal solution is chosen using a statistical analysis of the signal
24 by a cross-validation method. The depth where closed porosity over total porosity reaches 0.5
25 is 78.6 meters for Summit and 76 meters for NEEM. It means that ~50% of the gas becomes
26 unavailable for pumping at these depths because it is trapped in closed bubbles. Mean GEM
27 ages at these depths are 43.6 years for Summit and 67.5 years for NEEM.

28 **3.3 Hg physico-chemical behavior in the firn**

29 The calculations of GEM age distributions in the firn presented above and their use to
30 infer past atmospheric GEM variations implicitly assume that GEM concentrations in the firn
31 are not modified by chemical reactions or physical interaction with snow and ice surfaces.

1 While GEM is weakly adsorbed on ice surfaces (Bartels-Rausch et al., 2008), it can be
2 photochemically produced or oxidized in the seasonal snowpack (Lalonde et al., 2002;
3 Poulain et al., 2004). We cannot rule out that these reactions occur in the NEEM snowpack,
4 but their effects are probably limited to the top layers of the firn. Faïn et al. (2008) showed
5 that in central Greenland, these processes occur only during a limited period in summer. At
6 NEEM, the consistency of GEM concentrations at depths of 20, 30 and 40 m where the air is
7 < 5 years old (respectively 1.39, 1.34 and 1.39 ng.m⁻³) with the measured mean surface
8 atmospheric GEM concentration (1.38 ± 0.10 ng.m⁻³, n=327) suggests that the influence of
9 near-surface reactions on GEM is limited and within the measurement uncertainty. However,
10 long-term measurements of GEM in the atmosphere and in the snowpack and firn are needed
11 to firmly establish whether near-surface photochemical processes have any significant
12 influence on the GEM signal that is stored in the firn air.

13 **4 Results**

14 **4.1 GEM concentrations in firn air and in the atmosphere at NEEM**

15 GEM concentrations measured at 14 depths in the firn at NEEM are presented in Figure
16 2. Average GEM concentrations at these depths varied between 0.85 and 1.39 ng.m⁻³ with
17 maximum levels found in the upper layers and between 67 and 73 m. The mean surface
18 atmospheric GEM concentration was 1.38 ± 0.10 ng.m⁻³ (n=327) during the 6 days of
19 sampling. Figure 2 also shows measurements made both in firn air and in the atmosphere at
20 Summit Greenland in 2006 (Faïn et al., 2009).

21 Due to gravitational settling, gases having a higher atomic or molecular weight than air
22 show enrichment with depth in firn. This effect is enhanced for GEM due to its high atomic
23 weight. It can be illustrated by using a GEM constant concentration as input to the forward
24 firn model. On Figure 2, the (constant) atmospheric concentrations are adjusted in order to fit
25 the upper firn GEM data at Summit and NEEM. For Summit, an atmospheric level of 1.7
26 ng/m³ can fit the firn air data within uncertainties down to 66 m depth (corresponding to gas
27 ages between 1997 and 2006.4). For NEEM, an atmospheric GEM level of 1.38 ng/m³ can fit
28 the firn air data down to 50 m depth (corresponding to gas ages between 1997 and 2004.5).
29 The consistency of model results with upper firn GEM data at both sites using constant
30 atmospheric concentrations indicates no significant GEM trend over a decade preceding the
31 firn air pumping (see Table 1).

1 The low variability of GEM concentrations in the upper 30 meters of the firn also
2 suggests a small seasonal variation for atmospheric GEM. Measurements of atmospheric
3 GEM at Summit made over several weeks in spring and summer, confirm this (Brooks et al.,
4 2011)(Fain et al. personal communication). A large seasonal pattern (such as the one observed
5 at Alert) would induce steep concentration gradients in the upper 30 m meters of the firn, as
6 observed for CO and light alkanes (e.g. Helmig et al., 2014; Petrenko et al., 2013; Wang et
7 al., 2012). Below 40 m depth, firn air concentrations reflect an averaging of the atmospheric
8 signal over more than a year due to diffusional mixing.

9 The different atmospheric mean GEM concentrations (1.7 versus 1.38 ng/m³) at the two
10 sites however suggest a discrepancy between the two firn air pumping experiments and/or a
11 difference in the air masses reaching those sites.

12 **4.2 Single site atmospheric scenarios**

13 Figure 3 shows the optimal inverse model GEM scenarios that result in the fit with the
14 NEEM and Summit firn data, respectively. Different scenarios were tested to find the optimal
15 solution, as discussed in Wang et al. (2012) and these are presented in Figure S2. The root
16 mean square deviations between model results and data (RMSD_{model-data}) for the different
17 scenarios are provided in Table S1. Scenarios with the smoothest GEM variations induced
18 important mismatches between simulated and measured GEM profiles in firn, whereas
19 scenarios with the largest GEM variations required unrealistically large atmospheric
20 variations to match minor features in the data.

21 The optimal scenarios for Summit and NEEM show similar patterns with an increase of
22 atmospheric GEM concentration after the 1950's and an atmospheric GEM concentration
23 maximum in ~1970. However, the firn measurements and the single site scenario
24 reconstructions both indicate that there are inconsistencies between the Summit and NEEM
25 datasets. Compared to NEEM, the more pronounced peak shape of the Summit firn air GEM
26 in the 1970s requires larger atmospheric GEM variation in the corresponding optimal
27 scenario. Our optimal scenario for Summit is similar to the one obtained by Faïn et al. (2009)
28 using a Monte-Carlo method, although the ~1970 peak amplitude is sensitive to the choice of
29 the inverse model solution (Figure S2). In contrast, the optimal atmospheric GEM scenario
30 inferred from the NEEM firn air suggests only minor atmospheric GEM variations since 1950
31 when taking into account the experimental uncertainty.

1 **4.3 Multi-site scenario reconstructions**

2 Constraining the atmospheric trend with both Summit and NEEM data sets leads to a
3 poor match of the firm data (see upper panels of Figure 4). Consequently, $\text{RMSD}_{\text{model-data}}$
4 strongly increases (see Table 2) and leads to a large uncertainty envelope. Simple attempts to
5 reconcile the two records were made by subtracting a constant value from the Summit GEM
6 data and multiplying the Summit GEM data by constant value. The subtraction method led to
7 a minimum $\text{RMSD}_{\text{model-data}}$ value when decreasing the Summit data by 0.78 ng.m^{-3} (see Table
8 2 and Figure S3), but the resulting low upper firm values are likely inconsistent with
9 atmospheric GEM data. Better results, both in terms of $\text{RMSD}_{\text{model-data}}$ and Summit upper firm
10 results are obtained by multiplying the Summit GEM data by 0.6 (see Table 2 and lower
11 panels of Figure 4). This acceptable scenario is close to what we obtain using the single site
12 model on NEEM data (Figure 3). With this configuration, GEM peaks around 1970 ($1.42\text{-}1.96$
13 ng.m^{-3}) and would be minimal in 1998 (range $0.94\text{-}1.15 \text{ ng.m}^{-3}$). GEM levels in the 1940s
14 would be in the $0.55\text{-}1.06 \text{ ng.m}^{-3}$ range. However the origin of such a difference in GEM data
15 between Summit and NEEM remain to be explained. It is discussed in the following sections.

16 **5 Comparing the different factors influencing atmospheric GEM signals** 17 **recorded in the Arctic**

18 Possible explanations of the differences between NEEM and Summit data are geographical
19 variations, experimental bias, and atmospheric reactivity.

20 Atmospheric GEM measurements were conducted during different and short time periods and
21 the instrumental configuration differ from one site to another. Atmospheric GEM was $1.38 \pm$
22 0.10 ng.m^{-3} at NEEM 2009 during the field campaign while mean GEM was 1.77 ± 0.20
23 ng.m^{-3} at Summit in 2006 (Fain et al., 2009). Brooks et al. (2011) showed average values of
24 $1.31 \pm 0.21 \text{ ng.m}^{-3}$ (13 May – 15 June 2007) and of $1.45 \pm 0.11 \text{ ng.m}^{-3}$ (6 June – 17 July 2008)
25 at Summit. Lastly, a recent campaign at Summit from May 31 to August 2009 showed an
26 average GEM concentration of $1.35 \pm 0.13 \text{ ng.m}^{-3}$ (Fain et al, personal communication). As
27 discussed elsewhere (Slemr et al., 2015), GEM measurements made using different Tekran
28 2537A can show an average systematic uncertainty of 10 to 20%. From those partial data set,
29 measurements of GEM of Summit 2006 could be biased high as compared to other studies.

30 Relatively short-lived atmospheric species such as GEM may display large geographic
31 variations across the Arctic (Helmig et al., 2014). Summit ($72^{\circ}35''\text{N}$, $38^{\circ}28'\text{W}$, 3216 m
32 altitude) and NEEM (77.45°N 51.06°W , 2452 m altitude) are ~ 650 km distant and their

1 altitudes differ by more than 700 m, thus the dominant transport pathways of GEM to both
2 sites could differ. Atmospheric transport to Summit is different from High Arctic sites (Alert,
3 Zeppelin) due to its high altitude and remoteness from coastlines. Hirdman et al. (2010)
4 showed that the composition of air at Summit is not representative of the Arctic boundary
5 layer, but rather of the free troposphere. Summit is also strongly influenced by air transported
6 from North America, particularly in summer (Kahl et al., 1997). In contrast, Steen-Larsen et
7 al. (2011) suggested that air mass reaching NEEM originate primarily from the Arctic, the
8 Baffin Bay area (West of Greenland). More recently, continuous monitoring of the surface
9 water vapor isotopic composition at NEEM highlighted the predominance of Arctic air
10 masses and revealed some events where the moisture originates from the evaporation at the
11 Arctic sea-ice margin areas east of Greenland (Steen-Larsen et al., 2013). These finding
12 suggest that the NEEM GEM record is more reflective of Arctic air masses than the Summit
13 record, and differences between the two could be at least partly explained by contrasted
14 airsheds.

15 As suggested by the absence of seasonal variation of GEM in the upper firn at NEEM (see
16 section 4.1), NEEM site is however different from what is observed at Alert or Zeppelin.
17 NEEM is likely preserved from the influence of AMDEs in springtime and summer GEM
18 evasion from snow or the Arctic Ocean in summer.

19 This is an important result highlighting the benefit of conducting future studies in high
20 altitude sites in order to understand the GEM background signal of the Arctic troposphere. In
21 particular, year-round GEM measurements at the permanent Summit station during a few
22 years before a new firn air pumping would provide a way to cross check the mean annual
23 atmospheric GEM and concentrations at intermediate depths (~30-60 m) in the firn. Such
24 measurements would strongly improve our understanding of the link between GEM
25 concentrations at Arctic low-altitude sites and sites representative of free tropospheric
26 conditions.

27 **6 Implications for the atmospheric GEM history in the Arctic**

28 As shown on Figure 5, our optimal atmospheric GEM scenario is consistent with both
29 historical anthropogenic Hg emissions and atmospheric Hg concentrations simulated by a
30 fully coupled global biogeochemical box model presented in Horowitz et al. (2014).

31 Our optimal scenario show an increase of GEM concentrations during the 1950s and 1960s
32 which can be primarily related to the global rise in Hg emissions after the Second World War

1 (Faïn et al., 2009; Streets et al., 2011). At NEEM, GEM concentrations were maximal in
2 1965-1971 and minimal in 1995-1998. This is consistent with trends of global Hg releases to
3 air that show maxima of emissions to air in 1970 due to open-air waste burning and a
4 subsequent decrease due to environmental regulations in North America and Europe
5 (Horowitz et al., 2014).

6 The recent study of an ice-core drilled at NEEM in 2010 revealed a higher net rate of
7 atmospheric deposition of total Hg (the sum of all oxidized species of Hg, where GEM is
8 therefore not included) from late 1980s and late 1990s as compared to the onset of the 20th
9 century (Zheng, 2015). This higher deposition rate is also reflected in the Penny ice cap total
10 Hg record (Zdanowicz et al., 2015) in northern Canada on the west side of the Baffin bay. The
11 coincidence of the GEM minimum at NEEM and the maximum of total Hg deposition in the
12 ice-cores in the 1990s is striking in a context where global Hg emissions decreased from 1970
13 to 2000. Among several potential explanations, the declining sea-ice in this region and a
14 potential promotion of halogen radicals leading to an increased conversion of GEM to
15 depositable Hg species should be more investigated as proposed by others (Zdanowicz et al.,
16 2015). Although total Hg and GEM have different chemical properties, life-times and sources,
17 future studies should try to co-investigate those two fractions of atmospheric Hg in archive
18 records.

19 After the GEM minimum around 1995-1998, the optimal scenario suggests increasing
20 atmospheric GEM concentrations until 2010 that are consistent with the recent increase of
21 anthropogenic Hg releases in Asia and global tropospheric Hg simulations (Horowitz et al.,
22 2014). However, considering the experimental uncertainty and the adequate matching of the
23 upper firn data with constant atmospheric scenarios (see section 4.1 and Figure 2), this
24 optimal scenario should be taken with caution. A constant atmospheric GEM trend for the last
25 10 years preceding the firn record would be also a plausible alternative that is comprised
26 within the upper and upper bounds of scenarios in Figure 5.

27 Our result participates in the actual debate regarding recent atmospheric GEM trends. While
28 Horowitz et al. (2014) obtain an increase of atmospheric GEM at a global scale in response to
29 the recent increase of global Hg releases to air, a decrease of atmospheric GEM is observed
30 over the period 1995 to 2010 at Alert (Cole and Steffen, 2010; Cole et al., 2013). At Zeppelin,
31 atmospheric GEM do not show any clear trend over the period 2000-2009 (Berg et al., 2013;
32 Cole et al., 2013). Using the NEEM record, we cannot firmly conclude on recent GEM trends

1 in the Arctic. We can solely suggest that the sharp increase of the simulated global
2 tropospheric Hg using most recent global inventories and Hg releases to the air (Horowitz et
3 al., 2014) is not as clearly reflected in the GEM record at NEEM. It also suggests that
4 atmospheric GEM trends in the Arctic cannot be solely linked to anthropogenic emissions and
5 that the role of GEM sinks and natural sources have to be more carefully considered. For
6 instance, Chen et al. (2015) suggest a decline of Hg in the Arctic surface ocean over the
7 period 2000-2009. This decline may lead to decreasing natural GEM emissions from the
8 Arctic surface ocean although the influence of some environmental drivers (such as
9 temperature, sea-ice cover) should be taken into account. An increasing number of studies
10 suggest that sea-ice and the Hg cycle could be linked (Chen et al., 2015; Fisher et al., 2013;
11 Point et al., 2011; Stern et al., 2012; Zdanowicz et al., 2015). Therefore, the role of
12 climatological variables such as the sea-ice extent (coupled with the role of the ocean) should
13 be more deeply investigated in the future given the rapid changes of sea-ice extent observed in
14 the Arctic (Cavalieri and Parkinson, 2012).

15 Given the lack of current understanding of the Hg cycling in particular in polar regions, we
16 believe that predictions of future trends of atmospheric GEM concentration and Hg deposition
17 in those regions in response to changes in anthropogenic sources are highly hypothetical.

18 To answer these questions, an enhanced reinforced network of atmospheric Hg monitoring is
19 urgently needed in the Arctic and supplementary and long-term observations at high elevation
20 sites are needed.

21

22 **Acknowledgements**

23 This research was funded by the IPEV Program NEEM Mercury 1205. We thank our NEEM
24 collaborators for their assistance during the field campaign, especially Mauro Rubino of
25 CSIRO, Australia and the Centre for Ice and Climate, University of Copenhagen, now at the
26 Second University of Naples, Italy, Vas Petrenko of UCSD, USA, now at The University of
27 Rochester, USA, Zoe Courville of USACE, USA and Thomas Blunier (Niels Bohr Institute)
28 for their support. NEEM is directed and organized by the Center of Ice and Climate at the
29 Niels Bohr Institute and US NSF, Office of Polar Programs. It is supported by funding
30 agencies and institutions in Belgium (FNRS-CFB and FWO), Canada (GSC), China,
31 Denmark (FIST), France (IPEV, ANR, CEA and INSU/CNRS), Germany (AWI), Iceland
32 (RannIs), Japan (NIPR), Korea (KOPRI), The Netherlands (NWO/ALW), Sweden (VR),

1 Switzerland (SNF), United Kingdom (NERC) and the USA (US NSF, Office of Polar
2 Programs). AD acknowledges the Institut Universitaire de France. CSIRO's contribution was
3 supported in part by the Australian Climate Change Science Program, an Australian
4 Government Initiative. We gratefully acknowledge Sandy Steffen, Dorothy Durnford, David
5 G.Streets and Amanda Cole for sharing data and for the useful discussions.

6

7

1 7 References

2

3 Bartels-Rausch, T., Huthwelker, T., Jöri, M., Gäggeler, H.W., Ammann, M., 2008. Interaction
4 of gaseous elemental mercury with snow surfaces: laboratory investigation. *Environ. Res.*
5 *Lett.* 3, 045009.

6 Berg, T., Aspö, K., Steinnes, E., 2008. Transport of Hg from Atmospheric mercury
7 depletion events to the mainland of Norway and its possible influence on Hg deposition.
8 *Geophys. Res. Lett.* 35, L09802.

9 Berg, T., Pfaffhuber, K.A., Cole, A.S., Engelsen, O., Steffen, A., 2013. Ten-year trends in
10 atmospheric mercury concentrations, meteorological effects and climate variables at Zeppelin,
11 Ny-Alesund. *Atmos. Chem. Phys.* 13, 6575-6586.

12 Boyd, S., Vandenberghe, L., 2004. *Convex optimization*, Cambridge, New York.

13 Brooks, S., Moore, C., Lew, D., Lefer, B., Huey, G., Tanner, D., 2011. Temperature and
14 sunlight controls of mercury oxidation and deposition atop the Greenland ice sheet. *Atmos.*
15 *Chem. Phys.* 11, 8295-8306.

16 Buizert, C., Martinerie, P., Petrenko, V.V., Severinghaus, J.P., Trudinger, C.M., Witrant, E.,
17 Rosen, J.L., Orsi, A.J., Rubino, M., Etheridge, D.M., Steele, L.P., Hogan, C., Laube, J.C.,
18 Sturges, W.T., Levchenko, V.A., Smith, A.M., Levin, I., Conway, T.J., Dlugokencky, E.J.,
19 Lang, P.M., Kawamura, K., Jenk, T.M., White, J.W.C., Sowers, T., Schwander, J., Blunier,
20 T., 2012. Gas transport in firn: multiple-tracer characterisation and model intercomparison for
21 NEEM, Northern Greenland. *Atmos. Chem. Phys.* 12, 4259-4277.

22 Cavalieri, D.J., Parkinson, C.L., 2012. Arctic sea ice variability and trends, 1979-2010. *The*
23 *Cryosphere* 6, 881-889.

24 Chen, L., Zhang, Y., Jacob, D.J., Soerensen, A.L., Fisher, J.A., Horowitz, H.M., Corbitt, E.S.,
25 Wang, X., 2015. A decline in Arctic Ocean mercury suggested by differences in decadal
26 trends of atmospheric mercury between the Arctic and northern midlatitudes. *Geophys. Res.*
27 *Lett.* 42, 6076-6083.

28 Chen, N.H., Othmer, D.F., 1962. New Generalized Equation for Gas Diffusion Coefficient. *J.*
29 *Chem. Eng. Data* 7, 37-41.

30 Cole, A.S., Steffen, A., 2010. Trends in long-term gaseous mercury observations in the Arctic
31 and effects of temperature and other atmospheric conditions. *Atmos. Chem. Phys.* 10, 4661-
32 4672.

33 Cole, A.S., Steffen, A., Pfaffhuber, K.A., Berg, T., Pilote, M., Poissant, L., Tordon, R., Hung,
34 H., 2013. Ten-year trends of atmospheric mercury in the high Arctic compared to Canadian
35 sub-Arctic and mid-latitude sites. *Atmos. Chem. Phys.* 13, 1535-1545.

36 Dastoor, A.P., Durnford, D.A., 2013. Arctic Ocean: Is It a Sink or a Source of Atmospheric
37 Mercury? *Environ. Sci. Technol.* 48, 1707-1717.

38 Douglas, T.A., Loseto, L.L., Macdonald, R.W., Outridge, P., Dommergue, A., Poulain, A.,
39 Amyot, M., Barkay, T., Berg, T., Chételat, J., Constant, P., Evans, M., Ferrari, C., Gantner,
40 N., Johnson, M.S., Kirk, J., Kroer, N., Larose, C., Lean, D., Nielsen, T.G., Poissant, L.,
41 Rognerud, S., Skov, H., Sørensen, S., Wang, F., Wilson, S., Zdanowicz, C.M., 2012. The fate
42 of mercury in Arctic terrestrial and aquatic ecosystems, a review. *Environ. Chem.* 9, 321-355.

- 1 Durnford, D., Dastoor, A., 2011. The behavior of mercury in the cryosphere: A review of
2 what we know from observations. *J. Geophys. Res.-Atmos.* 116, D06305.
- 3 Durnford, D., Dastoor, A., Figueras-Nieto, D., Ryjkov, A., 2010. Long range transport of
4 mercury to the Arctic and across Canada. *Atmos. Chem. Phys.* 10, 6063-6086.
- 5 Faïn, X., Ferrari, C.P., Dommergue, A., Albert, M., Battle, M., Arnaud, L., Barnola, J.M.,
6 Cairns, W., Barbante, C., Boutron, C., 2008. Mercury in the snow and firn at Summit Station,
7 Central Greenland, and implications for the study of past atmospheric mercury levels. *Atmos.*
8 *Chem. Phys.* 8, 3441-3457.
- 9 Faïn, X., Ferrari, C.P., Dommergue, A., Albert, M.R., Battle, M., Severinghaus, J., Arnaud,
10 L., Barnola, J.-M., Cairns, W., Barbante, C., Boutron, C., 2009. Polar firn air reveals large-
11 scale impact of anthropogenic mercury emissions during the 1970s. *Proc. Natl. Acad. Sci. U.*
12 *S. A.* 106, 16114-16119.
- 13 Fisher, J.A., Jacob, D.J., Soerensen, A.L., Amos, H.M., Corbitt, E.S., Streets, D.G., Wang, Q.,
14 Yantosca, R.M., Sunderland, E.M., 2013. Factors driving mercury variability in the Arctic
15 atmosphere and ocean over the past 30 years. *Global Biogeochem. Cy.* 27, 1226-1235.
- 16 Fisher, J.A., Jacob, D.J., Soerensen, A.L., Amos, H.M., Steffen, A., Sunderland, E.M., 2012.
17 Riverine source of Arctic Ocean mercury inferred from atmospheric observations. *Nature*
18 *Geosci.* 5, 499-504.
- 19 Helmig, D., Petrenko, V., Martinerie, P., Witrant, E., Rockmann, T., Zuiderweg, A.,
20 Holzinger, R., Hueber, J., Thompson, C., White, J.W.C., Sturges, W., Baker, A., Blunier, T.,
21 Etheridge, D., Rubino, M., Tans, P., 2014. Reconstruction of Northern Hemisphere 1950-
22 2010 atmospheric non-methane hydrocarbons. *Atmos. Chem. Phys.* 14, 1463-1483.
- 23 Hirdman, D., Sodemann, H., Eckhardt, S., Burkhardt, J.F., Jefferson, A., Mefford, T., Quinn,
24 P.K., Sharma, S., Strom, J., Stohl, A., 2010. Source identification of short-lived air pollutants
25 in the Arctic using statistical analysis of measurement data and particle dispersion model
26 output. *Atmos. Chem. Phys.* 10, 669-693.
- 27 Horowitz, H.M., Jacob, D.J., Amos, H.M., Streets, D.G., Sunderland, E.M., 2014. Historical
28 Mercury releases from commercial products: global environmental implications. *Environ. Sci.*
29 *Technol.* 48, 10242-10250.
- 30 Kahl, J.D.W., Martinez, D.A., Kuhns, H., Davidson, C.I., Jaffrezo, J.-L., Harris, J.M., 1997.
31 Air mass trajectories to Summit, Greenland: A 44-year climatology and some episodic events.
32 *J Geophys Res-Oceans* 102, 26861-26875.
- 33 Lalonde, J.D., Poulain, A.J., Amyot, M., 2002. The role of mercury redox reactions in snow
34 on snow-to-air mercury transfer. *Environ. Sci. Technol.* 36, 174-178.
- 35 Larose, C., Dommergue, A., Maruszczak, N., Coves, J., Ferrari, C.P., Schneider, D., 2011.
36 Bioavailable Mercury Cycling in Polar Snowpacks. *Environ. Sci. Technol.* 45, 2150-2156.
- 37 Laube, J.C., Newland, M.J., Hogan, C., Brenninkmeijer, C.A.M., Fraser, P.J., Martinerie, P.,
38 Oram, D.E., Reeves, C.E., Rockmann, T., Schwander, J., Witrant, E., Sturges, W.T., 2014.
39 Newly detected ozone-depleting substances in the atmosphere. *Nature Geosci.* 7, 266-269.
- 40 Lugg, G.A., 1968. Diffusion coefficients of some organic and other vapors in air. *Anal.*
41 *Chem.* 40, 1072-&.
- 42 Massman, W.J., 1998. A review of the molecular diffusivities of H₂O, CO₂, CH₄, CO, O₃,
43 SO₂, NH₃, N₂O, NO, and NO₂ in air, O₂ and N₂ near STP. *Atmos. Environ.* 32, 1111-1127.

- 1 Massman, W.J., 1999. Molecular diffusivities of Hg vapor in air, O₂ and N₂ near STP and the
2 kinematic viscosity and thermal diffusivity of air near STP. *Atmos. Environ.* 33, 453–457.
- 3 Matsunaga, N., Hori, M., Nagashima, A., 1998. Diffusion coefficients of global warming
4 gases into air and its component gases. *High Temp. - High Pressures* 30, 77-83.
- 5 Petrenko, V.V., Martinerie, P., Novelli, P., Etheridge, D.M., Levin, I., Wang, Z., Blunier, T.,
6 Chappellaz, J., Kaiser, J., Lang, P., Steele, L.P., Hammer, S., Mak, J., Langenfelds, R.L.,
7 Schwander, J., Severinghaus, J.P., Witrant, E., Petron, G., Battle, M.O., Forster, G., Sturges,
8 W.T., Lamarque, J.F., Steffen, K., White, J.W.C., 2013. A 60 yr record of atmospheric carbon
9 monoxide reconstructed from Greenland firn air. *Atmos. Chem. Phys.* 13, 7567-7585.
- 10 Point, D., Sonke, J.E., Day, R.D., Roseneau, D.G., Hobson, K.A., Vander Pol, S.S., Moors,
11 A.J., Pugh, R.S., Donard, O.F.X., Becker, P.R., 2011. Methylmercury photodegradation
12 influenced by sea-ice cover in Arctic marine ecosystems. *Nature Geosci.* 4, 188-194.
- 13 Poulain, A.J., Lalonde, J.D., Amyot, M., Shead, J.A., Raofie, F., Ariya, P.A., 2004. Redox
14 transformations of mercury in an Arctic snowpack at springtime. *Atmos. Environ.* 38, 6763–
15 6774.
- 16 Rommelaere, V., Arnaud, L., Barnola, J.M., 1997. Reconstructing recent atmospheric trace
17 gas concentrations from polar firn and bubbly ice data by inverse methods. *J. Geophys. Res.-*
18 *Atmos.* 102, 30069-30083.
- 19 Schwander, J., Barnola, J.M., Andrie, C., Leuenberger, M., Ludin, A., Raynaud, D., Stauffer,
20 B., 1993. The age of the air in the firn and the ice at Summit, Greenland. *J. Geophys. Res.-*
21 *Atmos.* 98, 2831-2838.
- 22 Slemr, F., Angot, H., Dommergue, A., Magand, O., Barret, M., Weigelt, A., Ebinghaus, R.,
23 Brunke, E.G., Pfaffhuber, K.A., Edwards, G., Howard, D., Powell, J., Keywood, M., Wang,
24 F., 2015. Comparison of mercury concentrations measured at several sites in the Southern
25 Hemisphere. *Atmos. Chem. Phys.* 15, 3125-3133.
- 26 Soerensen, A.L., Jacob, D.J., Streets, D.G., Witt, M.L.I., Ebinghaus, R., Mason, R.P.,
27 Andersson, M., Sunderland, E.M., 2012. Multi-decadal decline of mercury in the North
28 Atlantic atmosphere explained by changing subsurface seawater concentrations. *Geophys.*
29 *Res. Lett.* 39.
- 30 Steen-Larsen, H.C., Johnsen, S.J., Masson-Delmotte, V., Stenni, B., Risi, C., Sodemann, H.,
31 Balslev-Clausen, D., Blunier, T., Dahl-Jensen, D., Ellehøj, M.D., Falourd, S., Grindsted, A.,
32 Gkinis, V., Jouzel, J., Popp, T., Sheldon, S., Simonsen, S.B., Sjolte, J., Steffensen, J.P.,
33 Sperlich, P., Sveinbjörnsdóttir, A.E., Vinther, B.M., White, J.W.C., 2013. Continuous
34 monitoring of summer surface water vapor isotopic composition above the Greenland Ice
35 Sheet. *Atmos. Chem. Phys.* 13, 4815-4828.
- 36 Steen-Larsen, H.C., Masson-Delmotte, V., Sjolte, J., Johnsen, S.J., Vinther, B.M., Bréon,
37 F.M., Clausen, H.B., Dahl-Jensen, D., Falourd, S., Fettweis, X., Gallée, H., Jouzel, J.,
38 Kageyama, M., Lerche, H., Minster, B., Picard, G., Punge, H.J., Risi, C., Salas, D.,
39 Schwander, J., Steffen, K., Sveinbjörnsdóttir, A.E., Svensson, A., White, J., 2011.
40 Understanding the climatic signal in the water stable isotope records from the NEEM shallow
41 firn/ice cores in northwest Greenland. *J. Geophys. Res.-Atm.* 116, D06108.
- 42 Steffen, A., Douglas, T., Amyot, M., Ariya, P., Aspmo, K., Berg, T., Bottenheim, J., Brooks,
43 S., Cobbett, F., Dastoor, A., Dommergue, A., Ebinghaus, R., Ferrari, C., Gardfeldt, K.,
44 Goodsite, M.E., Lean, D., Poulain, A.J., Scherz, C., Skov, H., Sommar, J., Temme, C., 2008.

1 A synthesis of atmospheric mercury depletion event chemistry in the atmosphere and snow.
2 Atmos. Chem. Phys. 8, 1445-1482.

3 Stern, G.A., Macdonald, R.W., Outridge, P.M., Wilson, S., Chetelat, J., Cole, A., Hintelmann,
4 H., Loseto, L.L., Steffen, A., Wang, F.Y., Zdanowicz, C., 2012. How does climate change
5 influence arctic mercury? Sci. Total Environ. 414, 22-42.

6 Streets, D.G., Devane, M.K., Lu, Z.F., Bond, T.C., Sunderland, E.M., Jacob, D.J., 2011. All-
7 Time Releases of Mercury to the Atmosphere from Human Activities. Environ. Sci. Technol.
8 45, 10485-10491.

9 Wang, Z., Chappellaz, J., Martinerie, P., Park, K., Petrenko, V., Witrant, E., Emmons, L.K.,
10 Blunier, T., Brenninkmeijer, C.A.M., Mak, J.E., 2012. The isotopic record of Northern
11 Hemisphere atmospheric carbon monoxide since 1950: implications for the CO budget.
12 Atmos. Chem. Phys. 12, 4365-4377.

13 Witrant, E., Martinerie, P., 2013. Input Estimation from Sparse Measurements in LPV
14 Systems and Isotopic Ratios in Polar Firns, in: Witrant, E., Martinez-Molina, J.J., Lovera, M.,
15 Senane, O., Dugard, L. (Eds.), 5th IFAC Symposium on System Structure and Control,
16 WTC, Grenoble, France, pp. 659–664.

17 Witrant, E., Martinerie, P., Hogan, C., Laube, J.C., Kawamura, K., Capron, E., Montzka,
18 S.A., Dlugokencky, E.J., Etheridge, D., Blunier, T., Sturges, W.T., 2012. A new multi-gas
19 constrained model of trace gas non-homogeneous transport in firn: evaluation and behaviour
20 at eleven polar sites. Atmos. Chem. Phys. 12, 11465-11483.

21 Zdanowicz, C., Krueemmel, E., Lean, D., Poulain, A., Kinnard, C., Yumvihoze, E., Chen, J.,
22 Hintelmann, H., 2015. Pre-industrial and recent (1970–2010) atmospheric deposition of
23 sulfate and mercury in snow on southern Baffin Island, Arctic Canada. Sci. Total Environ.
24 509–510, 104-114.

25 Zheng, J., 2015. Archives of total mercury reconstructed with ice and snow from Greenland
26 and the Canadian High Arctic. Sci. Total Environ. 509–510, 133-144.

27 Zuiderweg, A., Holzinger, R., Martinerie, P., Schneider, R., Kaiser, J., Witrant, E., Etheridge,
28 D., Petrenko, V., Blunier, T., Rockmann, T., 2013. Extreme C-13 depletion of CCl₂F₂ in firn
29 air samples from NEEM, Greenland. Atmos. Chem. Phys. 13, 599-609.

30
31

Table 1. Estimated mean age of GEM in firn air at Summit and NEEM for various depths

Summit			NEEM 2009		
Depth (m)	Mean Year	Age width	Depth (m)	Mean year	Age width
0	2006.4	0	0	2009.5	0
15	2004.8	2.4	20	2008.0	2.3
25	2003.4	5.2	30	2007.0	4.1
30	2002.7	6.3	40	2005.8	5.7
40	2001.3	8.2	50	2004.5	7.1
50	1999.9	9.6	60	2002.0	9.8
58	1998.7	10.6	62	2000.4	11.6
63	1997.8	11.3	64	1994.0	18.5
66	1997.0	12	67	1981.1	26.9
70	1994.0	15	69	1973.4	29.2
72	1987.2	20.8	72	1959.6	30
76	1972.8	25.5	73	1955.1	30.2
78	1965.1	27.1	74	1950.6	30.5
79.6	1959.6	27.7	75	1946.3	30.7
			76	1942.1	30.8

Table 2. Sensitivity of multi-site simulations to the assumed difference in atmospheric GEM between Summit and NEEM. The average mismatch between inverse model results and firm data (expressed as root mean square deviations in $\text{ng}\cdot\text{m}^{-3}$) is presented for various combinations. The RMSD for individual GEM measurements is $0.10 \text{ ng}\cdot\text{m}^{-3}$.

Multi-site simulations	RMSD	Multi-site simulations	RMSD
NEEM + 1 * Summit	0.35	NEEM + Summit-0	0.35
NEEM + 0.8*Summit	0.19	NEEM + Summit-0.38	0.22
NEEM + 0.7*Summit	0.13	NEEM + Summit-0.58	0.17
NEEM + 0.6*Summit	0.10	NEEM + Summit-0.78	0.15
NEEM + 0.5*Summit	0.14	NEEM + Summit-0.98	0.17

Figure 1



Figure 1: Map of the Arctic showing firn air measurements sites of NEEM and Summit. Measurement sites, where direct atmospheric GEM monitoring data are available, are also shown.

Figure 2

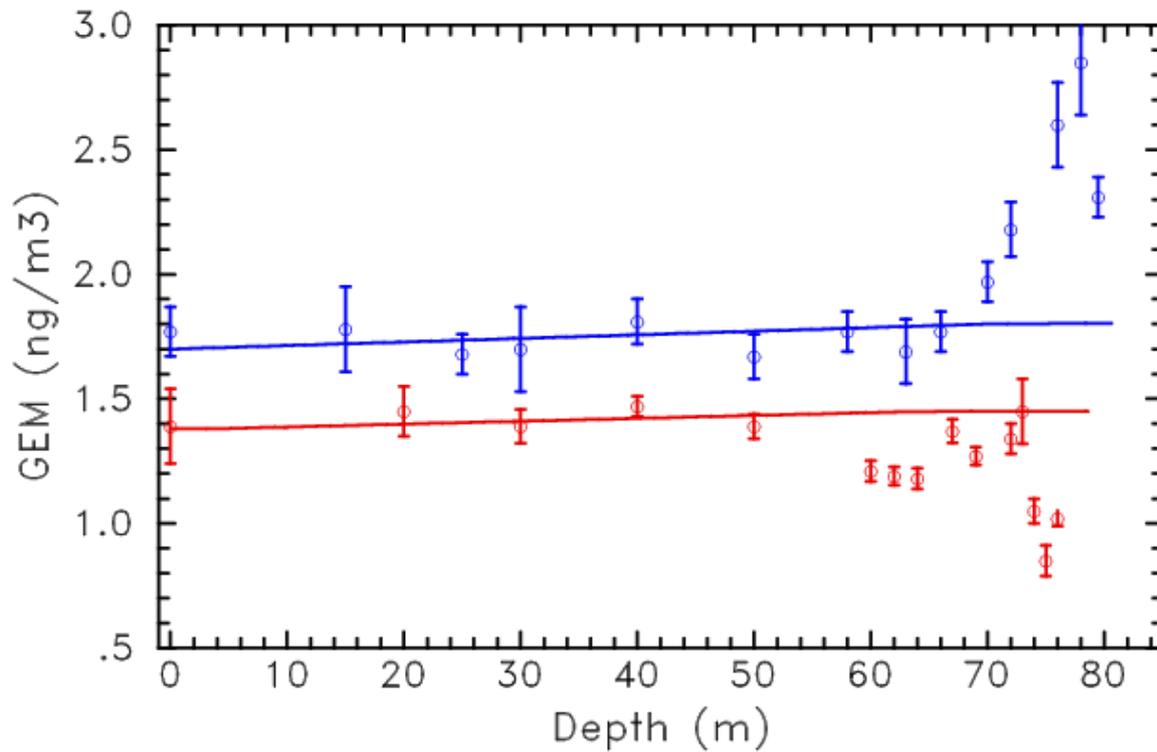


Figure 2: Mean GEM concentrations ($\pm\sigma$) measured in the firn at NEEM (red) and at Summit (blue). Summit data are from Faïn et al. (2009). Mean surface atmospheric GEM levels are also plotted. Constant atmospheric GEM levels of 1.7 ng.m^{-3} and 1.38 ng.m^{-3} are used to test the effect of GEM gravitational settling at Summit and NEEM respectively. In this case, GEM diffuses via gravity only and the corresponding GEM firn profiles that would be measured are shown by the blue and red line for Summit and NEEM respectively. These lines fit the firn measurements down to $\sim 50 \text{ m}$. It illustrates the fact that GEM concentrations have stayed roughly constant (given the experimental uncertainty) for the last decade preceding the firn pumping experiment.

Figure 3

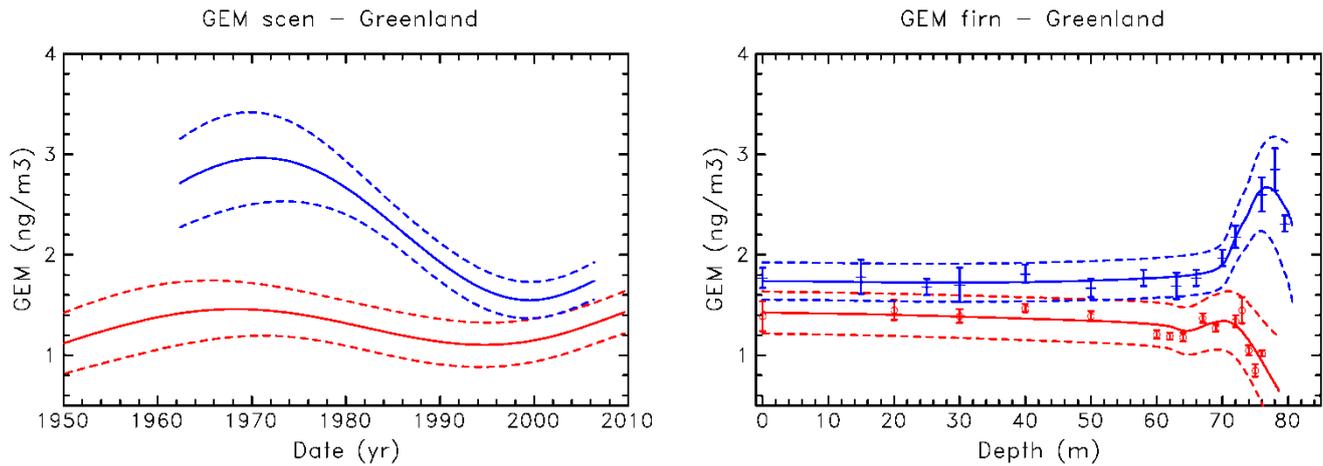


Figure 3: Left panel: scenarios of reconstructed atmospheric GEM using single site model at NEEM (red) and Summit (blue). The solid lines show the optimal scenario with its uncertainty envelopes as dashed lines. Other scenarios using different regularization terms are shown on supplementary figure S2. Right panel: GEM concentrations in firn obtained by the model using the corresponding scenarios of the left panels. Same color codes apply. The GEM firn air measurements are also reported.

Figure 4

Figure 4: Upper panels: optimal GEM scenarios using multi-site estimation. Left panels: best estimate time trends (continuous line) and uncertainty envelopes (dashed lines). Right panels: concentrations in Summit (blue line) and NEEM (red line) firns calculated with the optimal scenarios, compared with the measurements (circles with error bars). Optimal solutions using equal weights are used for both sites. Lower panels: same figure using a rescaling (multiplying factor) of 0.6 for Summit data.

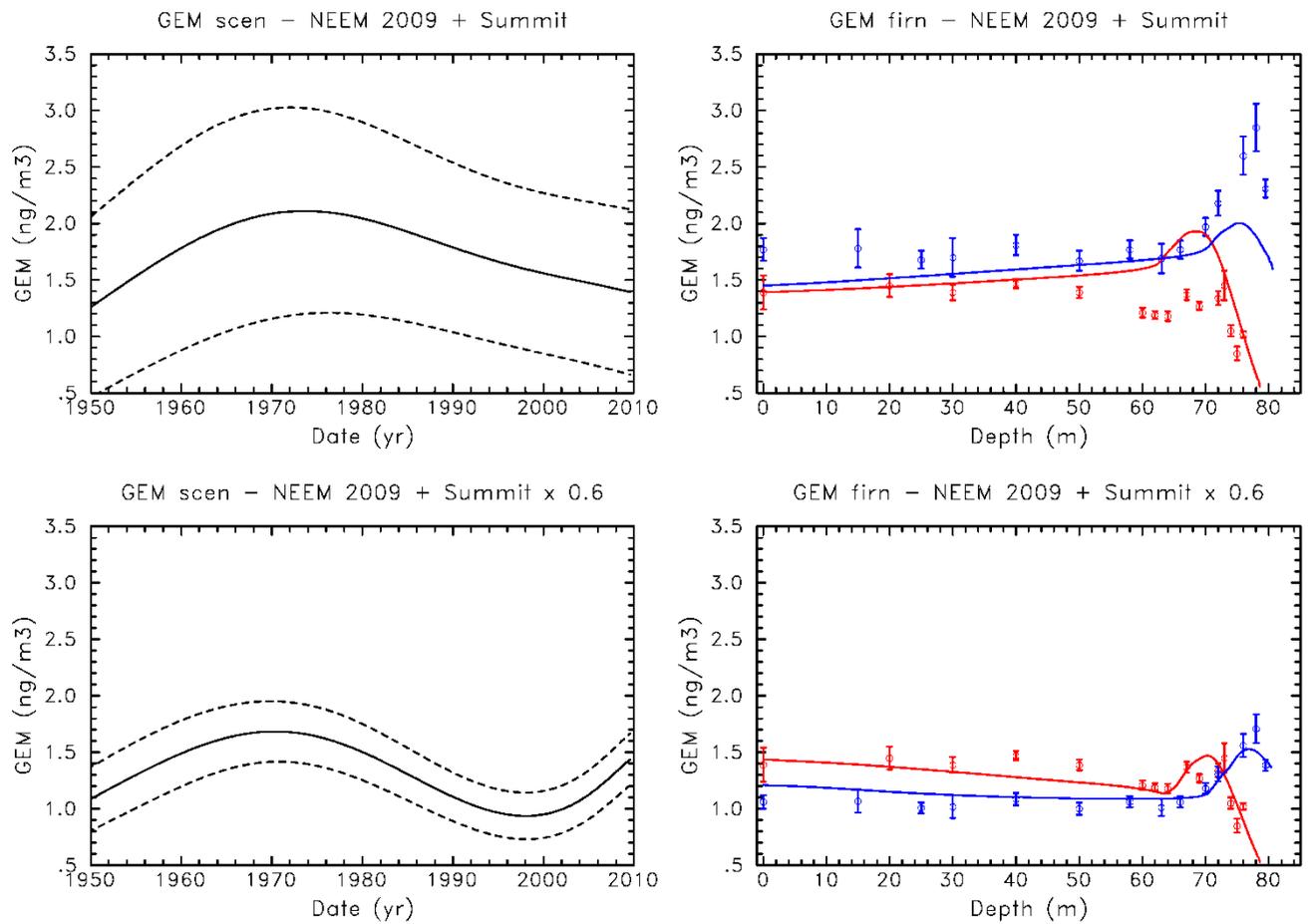
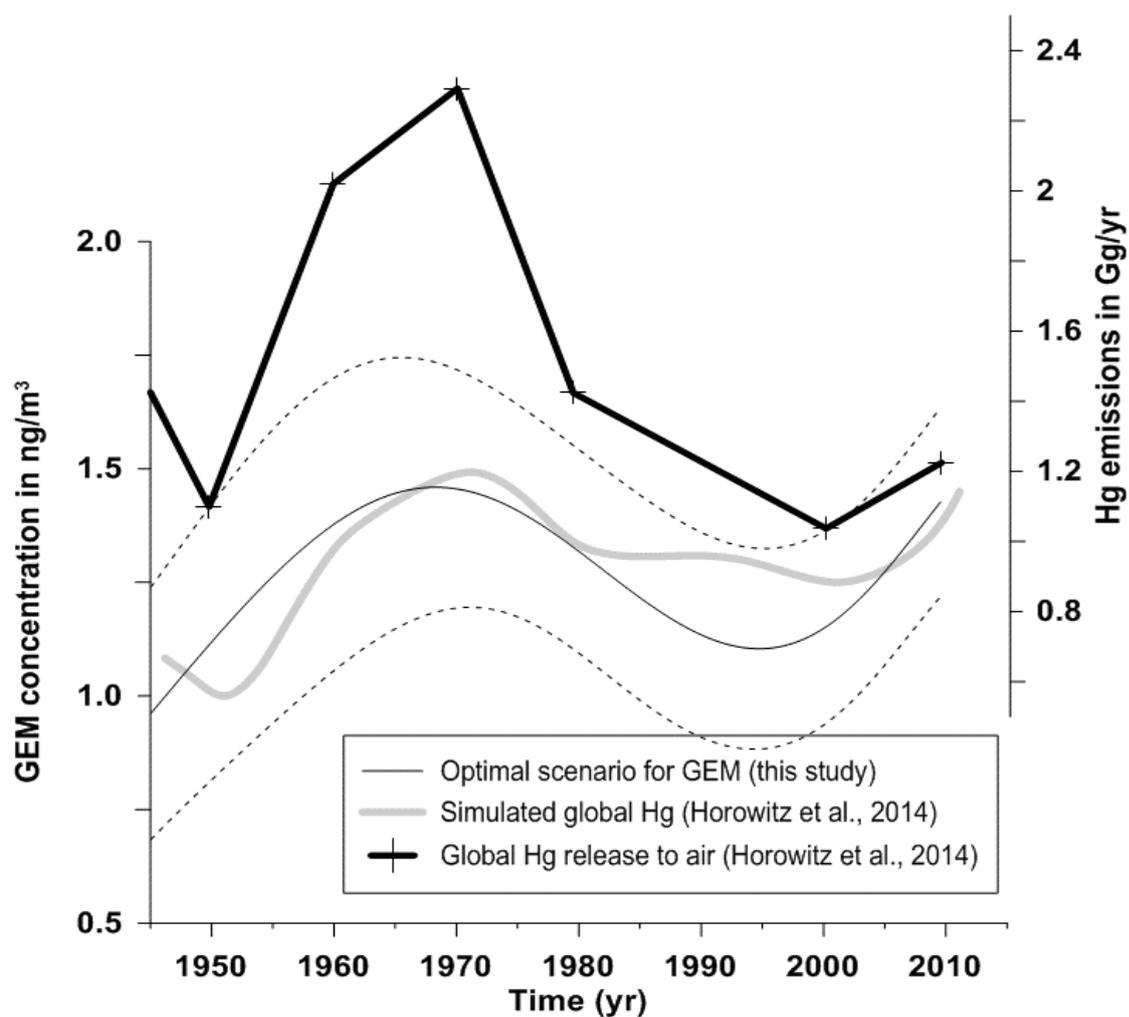


Figure 5



1

2

3 Figure 5: GEM trends as reconstructed by the firm air record. The black line with the two
4 dash lines represent the optimal scenario and its envelope obtained using NEEM data. The
5 gray line shows the simulated global tropospheric Hg signal and the bold black line represents
6 the history of global Hg releases to air. These data are extracted from Horowitz et al. (2014).

7